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(54) Title: HSD17B13 INHIBITORS



(57) **Abstract:** The present invention encompasses heteroaryl substituted 2,6-difluorophenol compounds of formula (I), wherein the groups A1 to A3, and Z have the meanings given in the claims and specification, their use in pharmaceutical compositions which contain these compounds and their use as medicaments, especially to interfere with the progression of liver disease from steatosis to later stages of nonalcoholic steatohepatitis, fibrosis, and cirrhosis.

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HSD17B13 INHIBITORS

Field of the invention

The present invention relates to heteroaryl substituted 2,6-difluorophenol compounds of formula (I),



wherein A_1 to A_3 , and Z have the meanings given in the claims and specification. Additionally disclosed are their use as inhibitors of HSD17B13, pharmaceutical compositions which contain said compounds and their use as medicaments, especially as agents for interfering with steatosis.

Background of the invention

WO 2021/211974 and WO 2022/020714 disclose thiophene-carboxamide HSD17B13 inhibitors.

WO 2022/020730 discloses quinazolinone HSD17B13 inhibitors.

HSD17B13 is a member of the 17b-hydroxysteroid dehydrogenases family of oxidoreductase enzymes that collectively act on a range of lipid substrates. In humans, HSD17B13 mRNA is most highly expressed in the liver, primarily in hepatocytes. Within the cell, HSD17B13 is associated with lipid droplets (Su et al, Proc National Acad Sci. 111: 11437–11442, 2014).

The physiological function of HSD17B13 is uncertain, and multiple substrates, including estradiol, retinol, and leukotriene B4, have been identified using an in vitro enzyme assay system in which NAD⁺ (nicotinamide adenine dinucleotide, oxidized form) acted as co-substrate (Abdul-Husn et al, The New England Journal of Medicine. 378: 1096–1106, 2018).

Loss of function (LoF) genetic variants in humans provide evidence for a role of HSD17B13 activity in mediating risk of certain liver diseases. The presence of the single nucleotide polymorphism (SNP) rs72613567 encoding a truncated, enzymatically inactive protein, has been associated with liver diseases such as liver fibrosis or non-alcoholic steatohepatitis (NASH).

The SNP rs72613567 SNP also mitigates the increased risk of liver disease.

The SNP rs72613567 was found to occur at a lower frequency in liver transplant recipients than in healthy controls.

The SNP rs62305723 encoding a HSD17B13 LoF variant has been associated with decreased severity of NASH (Ma et al, Hepatology 69: 1504–1519, 2018).

The SNP rs80182459 encoding a probable LoF variant has been found to be less frequent in certain patients with chronic liver disease (Kozlitina et al, The New England Journal of Medicine 379: 1876–1877, 2018).

SNP rs6834314 which is in high linkage with SNP rs72613567 was also found to be associated with fatty liver disease.

A hepatocyte-directed small interfering RNA (siRNA) designed to deplete HSD17B13 in human liver was found in 5 patients with fatty liver to decrease serum alanine aminotransferase (ALT) activity, a biomarker of liver damage.

In view of the data mentioned above it is desirable to provide potent HSD17B13 inhibitors.

According to the present invention, "HSD17B13 inhibitor(s)" means compounds which inhibit HSD17B13 in the test shown in examples 4 and 6.

Detailed description of the invention

Compounds of formula (I), wherein the groups A1 to A3 and Z have the meanings given hereinafter, were not known to act as HSD17B (17 β -Hydroxysteroid dehydrogenase) inhibitors selective for HSD17B13 as shown in example 12 by the comparative biochemical human IC₅₀ data for HSD17B11. Thus, the compounds according to the invention may be used for example for the treatment of steatosis such as non-alcoholic fatty liver disease (NAFLD) or non-alcoholic steatohepatitis (NASH). WO 2023/237504

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The present invention therefore relates to a compound of formula (I), or a salt thereof,







N NH



A skilled artisan is aware that several of heteroaryl groups for example can be described in form of different tautomers, i. e. pyrazoles, triazoles, imidazoles. Thus, the compounds of the present invention may exist as tautomeres. For example, any compound of the present invention which contains a pyrazole moiety as a heteroaryl group can exist as a 1H tautomer, or a 2H tautomer, or even a mixture in any amount of the two tautomers, or a triazole moiety can exist as a 1H tautomer, a 2H tautomer or a 4H tautomer, or even a mixture in any amount of said 1H, 2H or 4H tautomers, namely:





2H-tautomer







1H-tautomer

2H-tautomer

4H-tautomer

The present invention includes all possible tautomers of the compounds of the present invention as single tautomers, or as any mixture of said tautomers, in any ratio.





In one group of compounds according to the invention the structure of is selected from the group of structures consisting of





is

In yet another group of compounds of the invention the structure selected from the group of structures consisting of



The present invention is directed to compounds of formula (I) or salts thereof which interfer with lipogenesis wherein the selective inhibition of HSD17B13 is of therapeutic benefit, including but not limited to the treatment of non-alcoholic steatohepatitis. Thus, in another aspect the invention a compound of formula (I) or a pharmaceutically acceptable salt thereof is used as a medicament. The invention also relates to a compound of formula (I) or a pharmaceutically acceptable salt thereof for use in a method of treatment of the human or animal body.

Of particular interest are the use of one of the compounds of formula (I) or a salt thereof in the treatment of metabolic disorders. Therefore, another aspect of the invention is the use of a compound of formula (I) or a pharmaceutically acceptable salt thereof for preparing a pharmaceutical composition comprising at least one compound of formula (I) or a pharmaceutically acceptable salt thereof and a pharmaceutically acceptable carrier for the treatment of one of said disorders. Particularly preferred is their use in preparing a pharmaceutical composition to modulate metabolic disorders in the human or animal body.

Considered in the context of the present invention is also a method of treating a liver disease, metabolic disease, or cardiovascular disease using a compound disclosed herein, or a pharmaceutically acceptable salt, solvate or stereoisomer thereof, in combination with an additional therapeutic agent.

In some embodiments, the additional therapeutic agent is used for the treatment of diabetes or diabetes related disorder or conditions.

In some instances, the additional therapeutic agent comprises a statin, an insulin sensitizing drug, an insulin secretagogue, an alpha-glucosidase inhibitor, a GLP agonist, a THR beta agonist, a PDE inhibitor, a DPP-4 inhibitor (such as sitagliptin, vildagliptin, saxagliptin, linagliptin, anagliptin, teneligliptin, alogliptin, gemigliptin or dutogliptin) a catecholamine (such as epinephrine, norepinephrine or dopamine), a peroxisome-proliferator-activated receptor (PPAR)-gamma agonist (e.g. thiazolidinedione (TZD) [such as pioglitazone, rosiglitazone, rivoglitazone, or troglitazone], aleglitazar, farglitazar, muraglitazar or tesaglitazar, peroxisome-proliferator-activated receptor (PPAR)-alpha agonist, a peroxisome-proliferator-activated receptor (PPAR)-alpha agonist, a farnesoid X receptor (FXR) agonist (e.g. obeticholic acid), or a combination thereof.

In some cases, the statin is an HMG-CoA reductase inhibitor.

In other instances, additional therapeutic agents include fish oil, fibrate, vitamins such as niacin, retinoic acid (e. g. 9-cis retinoic acid), nicotinamide ribonucleoside or its analogs thereof, or combinations thereof. In other instances, additional therapeutic agents include

ACC inhibitors, FGF19 and FGF21 mimics, CCR3/CCR5 antagonists, or combinations thereof.

In some embodiments, the additional therapeutic agent is vivitrol.

In some embodiments, the additional therapeutic agent is a statin, such as an HMG-CoA reductase inhibitor, fish oil, fibrate, niacin, or a combination thereof. In other instances, the additional therapeutic agent is a dyslipidemia drug that prevents lipid absorption such as orlistat.

In some embodiments, the additional therapeutic agent is a vitamin such as retinoic acid or tocopheryl acetate for the treatment of diabetes and diabes related disorder or condition such as lowering elevated body weight and/or lowering elevated blood glucose from food intake. In some embodiments, the additional therapeutic agent is a glucose-lowering agent.

In some embodiments, the additional therapeutic agent is an anti-obesity agent.

In some embodiments, the additional therapeutic agent is selectected from among a peroxisome proliferator activated receptor (PPAR) agonist (gamma, dual or pan), a dipeptidyl peptidase (IV) inhibitor, a glucagon-like peptide-1 (GLP-1) analog, insulin or an insulin analog, an insulin secretagogue, a sodium glucose co-transporter 2 (SGLT2) inhibitor, a Glucophage, a human amylin analog, a biguanide, an alpha-glucosidase inhibitor, a meglitinide, a thiazolidinedione and sulfonylurea.

In some embodiments, the additional therapeutic agent is a lipid-lowering agent.

In some embodiments, the additional therapeutic agent is an antioxidant, corticosteroid such as budesonide, anti-tumor necrosis factor (TNF), or a combination thereof.

In some embodiments the additional therapeutic agent is administered at the same time as the compound disclosed herein.

In some embodiments, the additional therapeutic agent is administered less frequently than the compound disclosed herein.

In some embodiments, the additional therapeutic agent is administered more frequently than the compound disclosed herein.

In some embodiments, the additional therapeutic agent is administered prior than the administration of the compound disclosed herein.

In some embodiments, the additional therapeutic agent is administered after the administration of the compound disclosed herein.

In a treatment of a mammal, a compound according to the invention can be administered before, after or together with at least one other active substance or agent such as a diuretic, antihypertensive, lipid-lowering or antidiabetic agent.

FORMULATIONS

Suitable preparations for administering the compounds of the invention will be apparent to those with ordinary skill in the art and include for example tablets, pills, capsules, suppositories, lozenges, troches, solutions, elixirs, syrups, sachets, emulsions, inhalatives or dispersible powders. Preferred solutions are solutions for injection (s.c., i.v., i.m.) or solutions for infusion (injectables).

The content of the pharmaceutically active compound needs to be in amounts which are sufficient to achieve the dosage range specified below, for example in the range from 0.10 to 90 wt.-%, preferably 0.5 to 50 wt.-% of the composition as a whole. The doses specified may, if necessary, be given several times a day.

Suitable tablets may be obtained, for example, by mixing the active substance(s) of the invention with known excipients, for example inert diluents, carriers, disintegrants, adjuvants, surfactants, binders and/or lubricants.

Coated tablets may be prepared accordingly by coating cores produced analogously to the tablets with substances normally used for tablet coatings, for example collidone or shellac, gum arabic, talc, titanium dioxide or sugar. To achieve delayed release or prevent incompatibilities, the core may consist of multiple layers. Similarly, the tablet coating may consist of multiple layers to achieve delayed release, possibly using the excipients mentioned above for the tablets.

Syrups or elixirs containing the active substances or combinations thereof according to the invention may additionally contain a sweetener such as saccharine, cyclamate, glycerol or sugar and a flavour enhancer, *e. g.* a flavouring such as vanillin or orange extract. They may also contain suspension adjuvants or thickeners such as sodium carboxymethyl cellulose, wetting agents such as, for example, condensation products of fatty alcohols with ethylene oxide, or preservatives such as p-hydroxybenzoates.

Solutions for injection and infusion are prepared in the usual way, *e. g.* with the addition of isotonic agents, preservatives such as *p*-hydroxybenzoates, or stabilisers such as alkali metal salts of ethylenediamine tetraacetic acid, optionally using emulsifiers and/or dispersants, whilst if water is used as the diluent, for example, organic solvents may optionally be used as solvating agents or dissolving aids and transferred into injection vials or ampoules or infusion bottles.

Capsules may for example be prepared by mixing the active substance with an inert carrier such as lactose or sorbitol and packing them into gelatine capsules.

Suitable suppositories may be made for example by mixing with carriers provided for this purpose such as neutral fats or polyethyleneglycol or derivatives thereof.

Excipients which may be used include, for example, water, pharmaceutically acceptable organic solvents such as paraffins (*e.g.* petroleum fractions), vegetable oils (*e.g.* groundnut or sesame oil), mono- or polyfunctional alcohols (*e.g.* ethanol or glycerol), carriers such as *e.g.* natural mineral powders (*e.g.* kaolins, clays, talc, chalk), synthetic mineral powders (*e.g.* highly dispersed silicic acid and silicates), sugars (*e.g.* cane sugar, lactose and glucose), emulsifiers (*e.g.* lignin, spent sulphite liquors, methylcellulose, starch and polyvinylpyrrolidone) and lubricants (*e.g.* magnesium stearate, talc, stearic acid and sodium lauryl sulphate).

The preparations are administered by the usual methods, preferably by an oral or transdermal route, most preferably by oral route. For oral administration the tablets may of course contain, apart from the above-mentioned carriers, additives such as sodium citrate, calcium carbonate and dicalcium phosphate together with various additives such as starch, preferably potato starch, gelatine and the like. Moreover, lubricants such as magnesium stearate, sodium lauryl sulphate and talc may be used at the same time for the tabletting process. In aqueous suspensions the active substances may be combined with various flavour enhancers or colourings in addition to the excipients mentioned above.

For parenteral use, a solution of an active substance with suitable liquid carriers may be used.

The total amount of the active ingredient of formula (I) to be administered will generally range from about 0.001 mg/kg to about 200 mg/kg body weight per day, and preferably from about 0.01 mg/kg to about 20 mg/kg body weight per day. Clinically useful dosing schedules will range from one to three times a day dosing to once every four weeks dosing. In addition, "drug holidays" in which a patient is not dosed with a drug for a certain period of time, may be beneficial to the overall balance between pharmacological effect and tolerability. A unit dosage may contain from about 0.5 mg to about 1500 mg of active ingredient, and can be administered one or more times per day or less than once a day. The average daily dosage for administration by injection, including intravenous, intramuscular,

subcutaneous and parenteral injections, and use of infusion techniques will preferably be from 0.01 to 200 mg/kg of total body weight. The average daily rectal dosage regimen will preferably be from 0.01 to 200 mg/kg of total body weight. The average daily vaginal dosage regimen will preferably be from 0.01 to 200 mg/kg of total body weight. The average daily topical dosage regimen will preferably be from 0.1 to 200 mg administered between one to four times daily. The transdermal concentration will preferably be that required to maintain a daily dose of from 0.01 to 200 mg/kg. The average daily inhalation dosage regimen will preferably be from 0.01 to 100 mg/kg of total body weight.

It may sometimes be necessary to depart from the amounts specified, depending on a person's body weight, age, the route of administration, severity of the disease, individual response to a drug, nature of the formulation and the time or interval over which the drug is administered (continuous or intermittent treatment with one or multiple doses per day). Thus, in some cases it may be sufficient to use less than the minimum dose given above, whereas in other cases the upper limit may have to be exceeded. When administering a higher dosis, it may be advisable to divide the higher dosis into a number of smaller doses spread over the day.

"Pharmaceutically acceptable salts" as used herein refers to derivatives of the disclosed compounds wherein the parent compound is modified by making acid or base salts thereof. Examples of pharmaceutically acceptable salts include, but are not limited to, mineral or organic acid salts of basic residues such as amines; alkali or organic salts of acidic residues such as carboxylic acids; and the like.

For example, such salts include salts from benzenesulfonic acid, benzoic acid, citric acid, ethanesulfonic acid, fumaric acid, gentisic acid, hydrobromic acid, hydrochloric acid, maleic acid, malic acid, malonic acid, mandelic acid, methanesulfonic acid, 4-methyl-benzenesulfonic acid, phosphoric acid, salicylic acid, succinic acid, sulfuric acid and tartaric acid.

Further pharmaceutically acceptable salts can be formed with cations from ammonia, Larginine, calcium, 2,2'-iminobisethanol, L-lysine, magnesium, *N*-methyl-D-glucamine, potassium, sodium and tris(hydroxymethyl)-aminomethane.

The pharmaceutically acceptable salts of the present invention can be synthesized from the parent compound which contains a basic or acidic moiety by conventional chemical

methods. Generally, such salts can be prepared by reacting the free acid or base form of these compounds with a sufficient amount of the appropriate base or acid in water or in an organic diluent like ether, ethyl acetate, ethanol, isopropanol, or acetonitrile, or a mixture thereof.

Preparation of the compounds according to the invention

General

Unless stated otherwise, all the reactions are carried out in commercially obtainable apparatus using methods that are commonly used in chemical laboratories. Starting materials that are sensitive to air and/or moisture are stored under protective gas and corresponding reactions and manipulations therewith are carried out under protective gas (nitrogen or argon).

The compounds according to the invention are named in accordance with CAS rules using the software MarvinSketch (Chemaxon).

The compounds according to the invention are prepared by the methods of synthesis described hereinafter in which the substituents of the general formulae have the meanings given herein before. These methods are intended as an illustration of the invention without restricting its subject matter and the scope of the compounds claimed to these examples. Where the preparation of starting compounds is not described, they are commercially obtainable or may be prepared analogously to known compounds or methods described herein. Substances described in the literature are prepared according to the published methods of synthesis.

Overview

Synthesis of Compounds:

The compounds of the present invention can be prepared as discribed in the following section. The schemes and the procedures described below illustrate general synthetic routes to the compounds of general formula (I) of the invention and are not intended to be limiting. It is clear to the person skilled in the art that the order of transformations as exemplified in the schemes can be modified in various ways. The order of transformations exemplified in

the schemes is therefore not intended to be limiting. In addition, interconversion of any of the substituents can be achieved before and / or after the exemplified transformation. These modifications can be achieved by introduction of protecting groups, cleavage of protecting groups, exchange, reduction or oxidation of functional groups, halogenation, metallation, substitution or other reactions known to the person skilled in the art. These transformations include those which introduce a functionality which allows for further interconversion of substituents. Appropriate protecting groups and their introduction and cleavage are well-known to the person skilled in the art (see for example P.G.M. Wuts and T.W. Greene in "Protective Groups in Organic Synthesis", 4th edition, Wiley 2006). Specific examples are described in the subsequent paragraphs. Further, it is possible that two or more successive steps may be performed without work-up between said steps, e. g. in a "one-pot" reaction, as it is well-known to the person skilled in the art.

The synthesis of heteroaryl substituted 2,6-difluorophenol compounds according to the present invention are preferably carried out according to the general synthetic sequence, shown in **schemes 1-3**.



Scheme 1: Route for the preparation of compounds of the general formula **8** and **9**, wherein A1, A2 and A3 have the same meaning as given for the formula (I), supra, X has the meaning of Cl, Br, I, mesylate or triflate and Y has the meaning of Cl, Br or I and R has the meaning of alkyl.

Preparation of starting materials (Scheme 1):

The heterocyclic compounds of the general formulas **1**, **2**, **3**, **4**, **5**, **6** or 7 (Scheme 1) are commercially available or described in the literature.

Step 1 --> 2 (Scheme 1)

Halogenation Reaction

The conversion of compounds of the general formula 1 to compounds of the formula 2 is known to the skilled person. For Y=Br the reaction can be performed with reagents such as bromine, N-Bromosuccinimide or copper(ll) bromide. For Y=Cl the reaction can be performed with reagents such as N-chloro-succinimide or chlorine. For Y=I the reaction can be performed with reagents such as iodine or N-iodo-succinimide.

Step 2 --> 8 (Scheme 1)

Reduction of ester to alcohol

The conversion of compounds of the general formula **2** to alcohols of the formula **8** is known to the skilled person and can be performed with reagents such as lithium aluminium tetrahydride, sodium tetrahydridoborate, calcium borohydride, diisobutylaluminium hydride.

Step 3 --> 8 (Scheme 1)

Reduction of carboxylic acid to alcohol

The conversion of compounds of the general formula **3** to alcohols of the formula **8** is known to the skilled person and can be performed with reagents such as lithium aluminium tetrahydride.

Step 4 --> 8 (Scheme 1)

Reduction of aldehyde to alcohol

The conversion of compounds of the general formula **4** to alcohols of the formula **8** is known to the skilled person and can be performed with reagents such as sodium tetrahydroborate or lithium aluminium hydride.

Step 5 --> 6 (Scheme 1)

Metal-catalized methylation reaction

The conversion of compounds of the general formula **5** to compounds of the general formula **6** is known to the skilled person and can be performed for example under Suzuki conditions with reagents such as trimethylboroxin, a palladium catalyst such as (1,1'-bis(diphenylphosphino)ferrocene)palladium(II) dichloride or tetrakis(triphenylphosphine) palladium(0) and a base such as potassium carbonate in an organic solvent such as DMF or dioxane.

Step 7 --> 6 (Scheme 1) Halogenation Reaction

The conversion of compounds of the general formula 7 to compounds of the general formula 6 is known to the skilled person. For Y=Br the reaction can be performed with reagents such as bromine, N-Bromosuccinimide or copper(ll) bromide. For Y=Cl the reaction can be Fperformed with reagents such as N-chloro-succinimide or chlorine. For Y=I the reaction can be performed with reagents such as iodine or N-iodo-succinimide.

Step 6 --> 8 (Scheme 1)

Hydroxylation of methyl group

The conversion of compounds of the general formula **6** to alcohols of the formula **8** is known to the skilled person and can be performed with reagents such as selenium(IV) dioxide or *tert*-butylhydroperoxide/manganese triacetate.

Step 8 --> 9 (Scheme 1)

Conversion of hydroxy to halogen (Br, Cl, I), mesylate or triflate

The transformation of alcohols **8** to a halogenated compound of formula **9** can be performed (for X=Cl) for example using chlorinating reagents such as thionyl chloride, mesylchloride/triethylamine or triphenylphosphine/tetrachloromethane. For X=Br reagents such as phosphorus tribromide, trimethylsilyl bromide, hydrogen bromide, carbon

tetrabromide/triphenylphosphine or boron tribromide/triphenylphosphine can be used. For X=I reagents such as boron trifluoride diethyl etherate/potassium iodide, 1Himidazole/iodine/triphenylphosphine can be used. For X=mesylate reagents such as mesyl chloride and a base such as triethylamine can be used. For X=triflate reagents such as triflic anhydride and a base such as pyridine can be used.

Step 7 --> 9 (Scheme 1)

Radical halogenation of methyl group

The conversion of compounds of the general formula 7 to halides of the formula 9 is known to the skilled person. Typically, a radical starter such as dibenzoyl peroxide or 2,2'-azobis(isobutyronitrile) and a halogenation reagent are used. For X=Br the reaction can be performed with halogentation reagents such as N-Bromosuccinimide. For X=Cl the reaction can be performed with halogenation reagents such as N-chloro-succinimide.



Scheme 2: Route for the preparation of compounds of the general formulas **11, 12** and **14,** wherein R has the meaning of hydrogen or alkyl and R1 can be a protecting group known to the person skilled in the art (see for example T. W. Greene and P. G. M. Wuts in Protective Groups in Organic Synthesis, 3rd edition, Wiley 1999).

Preparation of starting materials (Scheme 2):

Compounds of the general formulas **10**, **11**, **12**, **13** or **14** (Scheme 2) are commercially available or described in the literature. 2,6-Difluorophenol is commercially available.

Step 2,6-Difluorophenol --> 10 (Scheme 2) Introduction of phenol protecting group

The phenol group of **2,6-difluorophenol** can be masked with a suitable protecing group R1 leading to compounds of the general formula **10**. The reaction conditions for introduction of such suitable protecting groups R1 are known to the skilled person (see for example Green, Wuts, "Protective groups in organic synthesis" 1999, John Wiley & Sons and references therein). Preferably benzyl, para-methoxybenzyl and 3,4-methoxybenzyl are used as protective groups during the synthesis.

Step 10 --> 11 (Scheme 2)

Introduction of boronic acid or boronic acid ester

The conversion of compounds of the general formula **10** to boronic acid derivatives of the formula **11** is known to the skilled person. For R=H the reaction can be performed with reagents such as triisopropyl borate and a base such as n-butyllithium followed by an acidic workup of the reaction. For R-R=-C(CH₃)₂-C(CH₃)₂- the reaction can be performed with reagents such as bis(pinacol)diborane and a catalyst such as Pt(N,N'-dicyclohexylimidazol-2-ylidene)(divinyltetramethylsiloxane).

Step 11 --> 12 (Scheme 2)

Removal of phenol protecting group

Removal of the protecting group R1 from compounds of formula **11** leads to compounds of formula **12**. The reaction conditions for removal of such suitable protecting groups R1 are known to the skilled person (see for example Green, Wuts, "Protective groups in organic synthesis" 1999, John Wiley & Sons and references therein). Preferably benzyl, paramethoxybenzyl and 3,4-methoxybenzyl are used as protective groups during the synthesis which can be removed by hydrogenation.

Step 13 --> 14 (Scheme 2)

Introduction of phenol protecting group

Compounds of the general formula **13** can be masked with a suitable protecing group R1 leading to compounds of the general formula **14**. The reaction conditions for introduction of such suitable protecting groups R1 are known to the skilled person (see for example Green, Wuts, "Protective groups in organic synthesis" 1999, John Wiley & Sons and references therein). Preferably benzyl, para-methoxybenzyl and 3,4-methoxybenzyl are used as protective groups during the synthesis.

Step 14 --> 11 (Scheme 2)

Introduction of boronic acid ester

The conversion of compounds of the general formula **14** to boronic acid derivatives of the formula **11** is known to the skilled person. For $R-R = -C(CH_3)_2-C(CH_3)_2$ - the reaction can be performed with reagents such as bis(pinacol)diborane, a catalyst such as palladium (II) [1,1'-bis(diphenylphosphanyl)ferrocene] dichloride and a base such as potassium acetate.

Step 13 --> 12 (Scheme 2)

Introduction of boronic acid ester

The conversion of compounds of the general formula **13** to boronic acid derivatives of the formula **12** is known to the skilled person. For $R-R = -C(CH_3)_2-C(CH_3)_2$ - the reaction can be performed with reagents such as bis(pinacol)diborane, a catalyst such as palladium (II) [1,1'-bis(diphenylphosphanyl)ferrocene] dichloride and a base such as potassium acetate.



Scheme 3: Route for the preparation of compounds of the general formula **18**, wherein A1, A2, A3 and Z have the same meaning as given for the formula (I), supra, X has the meaning of hydroxy, Cl, Br, I, mesylate or triflate, Y has the meaning of Cl, Br or I, R has the meaning of hydrogen or alkyl. In addition, R1 can be a a protecting group known to the person skilled in the art (see for example T. W. Greene and P. G. M. Wuts in Protective Groups in Organic Synthesis, 3rd edition, Wiley 1999).

Preparation of starting materials (Scheme 3):

Amide compounds of the general formulas **15** (Scheme 3) are commercially available, are described in the literature or can be prepared in analogy to literature procedures.

Step 8 + 15 --> 16 (Scheme 3)

Mitsunobu reaction

The conversion of compounds of the general formula **8** and compounds of the general formula **15** to derivatives of the formula **16** is known to the skilled person. The reaction can be performed under Mitsunobu conditions using reagents such as di-isopropyl azodicarboxylate/triphenylphosphine, di-tert-butyl azodicarboxylate/triphenylphosphine or dietylazodicarboxylate/ triphenylphosphine.

Step 9 + 15 --> 16 (Scheme 3)

Alkylation of amide

The conversion of compounds of the general formula **9** and compounds of the general formula **15** to derivatives of the formula **16** is known to the skilled person. For X=Cl, Br or I the reaction can be performed with a base such as potassium carbonate, caesium carbonate, sodium hydride or LDA in solvents such as DMF, DMSO or acetonitrile.

Step 16 + 11 --> 17 (Scheme 3)

Palladium catalyzed reaction with boronic acids

Heteroaryl halides of formula **16** can be reacted with a boronic acid derivative **11** to give a compound of formula **17**. The boronic acid derivative may be a boronic acid (R=H) or an ester of the boronic acid, e.g. its iropropyl ester (R = $-CH(CH_3)_2$), preferably an ester derived from pinacol in which the boronic acid intermediate forms a 2-aryl-4,4,5,5-tretramethyl-1,3,2,-dioxoborolane (R-R = $-C(CH_3)_2-C(CH_3)_2-$).

The coupling reaction is catalyzed by palladium catalysts, e.g. by a Pd (0) catalyst like tetrakis(triphenylphosphine)palladium (0) [Pd (PPh₃)₄], tris(dibenzylidideneacetone)dipalladium (0) [Pd₂(dba)₃], or by Pd (II) catalysts like dichlorobis(triphenylphosphine)-palladium (II) [Pd (PPh₃)₂Cl₂], XPhos Pd G2, Pd-Peppsi 2Me-Ipent Cl, palladium (II) acetate and triphenylphosphine or by [1,1`-bis(diphenylphosphine)ferrocene] palladium dichloride.

The reaction is preferably carried out in a mixture of a solvent like 1,2-dimethoxymethane, dioxane, DMF, DME, THF, ethanol or isopropanol with water and in the presence of a base like potassium carbonate, sodium bicarbonate, potassium acetate or potassium phosphate. (review: D.G. Hall, Boronic Acids, 2005 WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim, ISBN 3-527-30991-8 and references therein).

The reaction is performed at temperatures ranging from room temperature (i.e. approx. 20°C) to the boiling point of the respective solvent. Further on, the reaction can be performed at temperatures above the boiling point using pressure tubes and a microwave oven. The reaction is preferably completed after 1 to 36 hours of reaction time.

Step 17 --> 18 (Scheme 3)

Removal of phenol protecting group

Removal of the protecting group R1 from compounds of formula 17 leads to compounds of formula 18. The reaction conditions for removal of such suitable protecting groups R1 are known to the skilled person (see for example Green, Wuts, "Protective groups in organic synthesis" 1999, John Wiley & Sons and references therein). Preferably benzyl, para-methoxybenzyl and 3,4-methoxybenzyl are used as protective groups during the synthesis which can be removed by hydrogenation.

Step 16 --> 19 (Scheme 3)

Introduction of boronic acid ester

The conversion of compounds of the general formula **16** to boronic acid derivatives of the formula **19** is known to the skilled person. For $R-R=-C(CH_3)_2-C(CH_3)_2$ - the reaction can be performed with reagents such as bis(pinacol)diborane, a catalyst such as palladium (II) [1,1'-bis(diphenylphosphanyl)ferrocene] dichloride and a base such as potassium acetate.

Step 19 + 13--> 18 (Scheme 3)

Palladium catalyzed reaction with boronic acids

Aryl halides of formula **13** can be reacted with a boronic acid derivative **19** to give a compound of formula **18**. The boronic acid derivative may be a boronic acid (R=H) or an ester of the boronic acid, e.g. its iropropyl ester (R=-CH(CH₃)₂), preferably an ester derived from pinacol in which the boronic acid intermediate forms a 2-aryl-4,4,5,5-tretramethyl-1,3,2,-dioxoborolane (R-R = -C(CH₃)₂-C(CH₃)₂-).

The coupling reaction is catalyzed by palladium catalysts, e.g. by Pd (0) catalyst like tetrakis(triphenylphosphine)palladium (0) [Pd (PPh₃)₄], tris(dibenzylidideneacetone)dipalladium (0) [Pd₂(dba)₃], or by Pd (II) catalysts like dichlorobis(triphenylphosphine)-palladium (II) [Pd (PPh₃)₂Cl₂], XPhos Pd G2, Pd-Peppsi 2Me-Ipent Cl, palladium (II) acetate and triphenylphosphine or by [1,1`-bis(diphenylphosphine)ferrocene]palladium dichloride.

The reaction is preferably carried out in a mixture of a solvent like 1,2-dimethoxymethane, dioxane, DMF, DME, THF, ethanol or isopropanol with water and in the presence of a base like potassium carbonate, sodium bicarbonate, potassium acetate or potassium phosphate (review: D.G. Hall, Boronic Acids, 2005 WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim, ISBN 3-527-30991-8 and references therein).

The reaction is performed at temperatures ranging from room temperature (i.e. approx. 20°C) to the boiling point of the respective solvent. Further on, the reaction can be performed at temperatures above the boiling point using pressure tubes and a microwave oven. The reaction is preferably completed after 1 to 36 hours of reaction time.

Step 16 + 12 --> 18 (Scheme 3)

Palladium catalyzed reaction with boronic acids

Heteroaryl halides of formula **16** can be reacted with a boronic acid derivative **12** to give a compound of formula **18**. The boronic acid derivative may be a boronic acid (R=H) or an ester of the boronic acid, e.g. its iropropyl ester (R = $-CH(CH_3)_2$), preferably an ester derived from pinacol in which the boronic acid intermediate forms a 2-aryl-4,4,5,5-tretramethyl-1,3,2,-dioxoborolane (R-R = $-C(CH_3)_2-C(CH_3)_2$ -).

The coupling reaction is catalyzed by palladium catalysts, e.g. by Pd (0) catalyst like tetrakis(triphenylphosphine)palladium (0) [Pd (PPh₃)₄], tris(dibenzylidideneacetone)dipalladium (0) [Pd₂(dba)₃], or by Pd (II) catalysts like dichlorobis(triphenylphosphine)-palladium (II) [Pd (PPh₃)₂Cl₂], XPhos Pd G2, Pd-Peppsi 2Me-Ipent Cl, palladium (II) acetate and triphenylphosphine or by [1,1`-bis(diphenylphosphino)ferrocene] palladium dichloride.

The reaction is preferably carried out in a mixture of a solvent like 1,2-dimethoxymethane, dioxane, DMF, DME, THF, ethanol or isopropanol with water and in the presence of a base like potassium carbonate, sodium bicarbonate, potassium acetate or potassium phosphate (as reviewed in D.G. Hall, Boronic Acids, 2005 WILEY-VCH Verlag GmbH & Co. KGaA,

Weinheim, ISBN 3-527-30991-8 and references therein).

The reaction is performed at temperatures ranging from room temperature (i.e. approx. 20°C) to the boiling point of the respective solvent. Further on, the reaction can be performed at temperatures above the boiling point using pressure tubes and a microwave oven. The reaction is preferably completed after 1 to 36 hours of reaction time.

Step 19 + 14 --> 17 (Scheme 3)

Palladium catalyzed reaction with boronic acids

Aryl halides of formula 14 can be reacted with a boronic acid derivative 19 to give a compound of formula 17. The boronic acid derivative may be a boronic acid (R=H) or an ester of the boronic acid, e. g. its iropropyl ester (R=-CH(CH₃)₂), preferably an ester derived from pinacol in which the boronic acid intermediate forms a 2-aryl-4,4,5,5-tretramethyl-1,3,2-dioxoborolane (R-R=-C(CH₃)₂-C(CH₃)₂-).

The coupling reaction is catalyzed by palladium catalysts, e.g. by Pd (0) catalyst like tetrakis(triphenylphosphine)palladium (0) [Pd (PPh₃)₄], tris(dibenzylidideneacetone)dipalladium (0) [Pd₂(dba)₃], or by Pd (II) catalysts like dichlorobis(triphenylphosphine)-palladium (II) [Pd (PPh₃)₂Cl₂], XPhos Pd G2, Pd-Peppsi 2Me-Ipent Cl, palladium (II) acetate and triphenylphosphine or by [1,1[°]-bis(diphenylphosphine)ferrocene] palladium dichloride.

The reaction is preferably carried out in a mixture of a solvent like 1,2-dimethoxymethane, dioxane, DMF, DME, THF, ethanol or isopropanol with water and in the presence of a base like potassium carbonate, sodium bicarbonate, potassium acetate or potassium phosphate (review: D.G. Hall, Boronic Acids, 2005 WILEY-VCH Verlag GmbH & Co. KGaA, Weinheim, ISBN 3-527-30991-8 and references therein).

The reaction is performed at temperatures ranging from room temperature (i.e. approx. 20°C) to the boiling point of the respective solvent. Further on, the reaction can be performed at temperatures above the boiling point using pressure tubes and a microwave oven. The reaction is preferably completed after 1 to 36 hours of reaction time.

General preparation method for compounds of formula (I):

The present invention discloses the following method or process to prepare compounds of general formula (I) using compounds X1 and X2:



Said method to obtain compounds of general formula (I) is characterized in that

- In the compounds of general formula X1, Z, A1, A2 and A3 have the same meaning as defined for the compounds of the general formula (I),
- the compounds of general formula X1, wherein Y can be Br, I or Cl, are reacted with a compound of general formula X2 using a palladium catalyst and a base; and
- the reaction takes place in a solvent or solvent mixture at a temperature between ambient temperatur and the boiling point of the solvent, preferably between 50°C and 120°C.
- The preparation of the compounds of the general formula (I) can be performed in an aprotic or protic solvent or a solvent mixture, preferably in 1,4-dioxane, tetrahydrofurane, or ethanol/water.
- Preferred bases which can be used for the preparation of compounds of the general formula (I) are sodium carbonate or cesium carbonate.
- Preferred palladium catalysts for preparation of compounds of general formula (I) are
 - chloro(2-dicyclohexylphosphino-2',4',6'-triisopropyl-1,1'-biphenyl) [2-(2'-amino-1,1'-biphenyl)] palladium (II) (2nd Generation XPhos Precatalyst) or
 - 1,3-Bis(2,6-Di-3-pentylphenyl) imidazol-2-ylidene] (3-chloropyridyl) dichloropalladium (II).

Intermediate compounds:

The present invention also discloses intermediate compounds useful in the preparation of compounds of general formula (I). In particular, the invention discloses compounds of general formula Ia in which Z is as defined for the compounds of general formula (I) *supra* and Y can be Br, I or Cl:



The present invention discloses further intermediate compounds useful in the preparation of compounds of formula (I). In particular, the present invention also discloses compounds of formula (Ib) in which in which A1, A2 and A3 are as defined for the compounds of general formula (I) *supra*:





The following Table lists compounds 01-30 prepared according to the present invention.














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Features and advantages of the present invention will become apparent from the following examples which illustrate the invention by way of example without restricting its scope.

ADDREVIA	10115
ACN	acetonitrile
АсОН	acetic acid
aq.	aqueous
°C	degree Celsius
CDI	1,1'-carbonyldiimidazole
conc.	concentrated
Cs ₂ CO ₃	cesium carbonate
DCM	dichloromethane
DIPEA	N,N-diisopropylethylamine
DMF	N,N-dimethylformamide
DTAD	di- <i>tert</i> butyl azodicarboxylate
EDC	({[3-(dimethylamino)propyl]imino}methylidene)(ethyl)amine
ESI-MS	Electrospray ionization mass spectrometry
Et ₂ O	diethyl ether
EtOAc	ethyl acetate
EtOH	ethanol
h	hour
HCl	hydrochloric acid
HOBT	1-hydroxybenzotriazole
HPLC	High performance liquid chromatography
HMDS	bis(trimethylsilyl)amine
K ₂ CO ₃	potassium carbonate
KOAc	potassium acetate
KOtBu	potassium <i>tert</i> butoxide
K ₃ PO ₄	tripotassium phosphate
L	liter
LiAlH ₄	lithium aluminium hydride
МеОН	methanol
L	

ABBREVIATIONS

minminutemLmillilitern-BuLin-butyllithiumNaOAcsodium acetateNa2SO4sodium sulfateNaOC1sodium hypochloriteNaOHsodium hydroxidePdCl2(dtbpf)[1,1'-bis(di-tert-butylphosphino)ferrocene]dichloropalladium(IIPd(dppf)2 x1,1'-bis(diphenyl-phosphino)-ferrocen-palladium(II) dichloride
n-BuLi <i>n</i> -butyllithiumNaOAcsodium acetateNa2SO4sodium sulfateNaOClsodium hypochloriteNaOHsodium hydroxidePdCl2(dtbpf)[1,1'-bis(di-tert-butylphosphino)ferrocene]dichloropalladium(IIPd(dppf)2 x1,1'-bis(diphenyl-phosphino)-ferrocen-palladium(II) dichloride
NaOAcsodium acetateNa2SO4sodium sulfateNaOC1sodium hypochloriteNaOHsodium hydroxidePdCl2(dtbpf)[1,1'-bis(di-tert-butylphosphino)ferrocene]dichloropalladium(IIPd(dppf)2 x1,1'-bis(diphenyl-phosphino)-ferrocen-palladium(II) dichloride
Na2SO4sodium sulfateNaOC1sodium hypochloriteNaOHsodium hydroxidePdCl2(dtbpf)[1,1'-bis(di-tert-butylphosphino)ferrocene]dichloropalladium(IIPd(dppf)2 x1,1'-bis(diphenyl-phosphino)-ferrocen-palladium(II) dichloride
NaOClsodium hypochloriteNaOHsodium hydroxidePdCl2(dtbpf)[1,1'-bis(di-tert-butylphosphino)ferrocene]dichloropalladium(IIPd(dppf)2 x1,1'-bis(diphenyl-phosphino)-ferrocen-palladium(II) dichloride
NaOHsodium hydroxidePdCl2(dtbpf)[1,1'-bis(di-tert-butylphosphino)ferrocene]dichloropalladium(IIPd(dppf)2 x1,1'-bis(diphenyl-phosphino)-ferrocen-palladium(II) dichloride
PdCl2(dtbpf)[1,1'-bis(di-tert-butylphosphino)ferrocene]dichloropalladium(IIPd(dppf)2 x1,1'-bis(diphenyl-phosphino)-ferrocen-palladium(II) dichloride
Pd(dppf) ₂ x 1,1'-bis(diphenyl-phosphino)-ferrocen-palladium(II) dichloride
CH ₂ Cl ₂ dichloromethane complex
Pd-Peppsi [1,3-bis[2,6-bis(1-ethylpropyl)phenyl]-4,5-dichloro-imidazol-2-yl]
2Me Ipent Cl dichloro-(2-methyl-1-pyridyl)palladium
RT room temperature (about 20°C)
sat. saturated
t-BuOH tert-butanol
TEA triethylamine
TFA trifluoroacetic acid
THF tetrahydrofuran
TPP triphenylphosphine
XPhos2-(dicyclohexylphosphanyl)-2',4',6'-tris(isopropyl)biphenyl
XPhos Pd G2 chloro(2-dicyclohexylphosphino-2',4',6'-tri-i-propyl-1,1'-biphenyl)
(2'-amino-1,1'-biphenyl-2-yl) palladium (II)

EXAMPLES

Example 1: Analytical HPLC methods

1.1 Method A

Analytical column: Waters; Sunfire C18_3.0 x 30 mm, 2.5 μ m; column temperature: 60° Device: Agilent 1200 with DA- and MS-Detector

time (min)	Vol% water	Vol% ACN	Flow
	(incl. 0.1% TFA)		[mL/min]
0.00	97.0	3.0	2.2
0.20	97.0	3.0	2.2
1.20	0.0	100.0	2.2
1.25	0.0	100.0	3.0
1.40	0.0	100.0	3.0

1.2 Method B

Analytical column: Waters; XBridge C18_3.0 x 30 mm, 2.5 μ m; column temperature: 60° Device: Agilent 1200 with DA- and MS-Detector

time (min)	Vol% water	Vol% ACN	Flow
	(incl. 0.1% NH4OH)		[mL/min]
0.00	97.0	3.0	2.2
0.20	97.0	3.0	2.2
1.20	0.0	100.0	2.2
1.25	0.0	100.0	3.0
1.40	0.0	100.0	3.0

1.3 Method C

Analytical column: Waters; Sunfire C18_2.1 x 30 mm, 2.5 μ m; column temperature: 60° Device: Waters Acquity with DA- and MS-Detector

time (min)	Vol% water	Vol% ACN	Flow
	(incl. 0.1% TFA)		[mL/min]
0.00	99.0	1.0	1.5
0.02	99.0	1.0	1.5
1.00	0.0	100.0	1.5
1.10	0.0	100.0	1.5

1.4 Method D

Analytical column: Sunfire C18_3.0 x 30 mm, 2.5 μ m; column temperature: 60° Device: Waters Acquity, QDa Detector

time (min)	Vol% water	Vol% ACN	Flow
	(incl. 0.1% TFA)	(incl. 0.08% TFA)	[mL/min]
0.00	95.0	5.0	1.5
1.30	0.0	100.0	1.5
1.50	0.0	100.0	1.5
1.60	95.0	5.0	1.5

1.5 Method E

Analytical column: Sunfire C18_3.0 x 30 mm, 2.5 μ m; column temperature: 60° Device: Waters Acquity, QDa Detector

time (min)	Vol% water	Vol% ACN	Flow
	(incl. 0.1% TFA)	(incl. 0.08% TFA)	[mL/min]
0.00	95.0	5.0	1.5
1.30	0.0	100.0	1.5
1.50	0.0	100.0	1.5
1.60	95.0	5.0	1.5

Example 2: Preparation of Intermediates

2.1 Intermediate I

Intermediate I.1 (general procedure)

1-[(2-Bromo-1,3-thiazol-5-yl)methyl]-3-methyl-1,2,3,4-tetrahydropyrimidine-2,4-dione



3-Methyl-1,2,3,4-tetrahydropyrimidine-2,4-dione (3.00 g; 23.79 mmol) and 3-bromothiazol-5-methanol (5.08 g; 26.17 mmol) were dissolved in THF (50 mL) and DMF (20 mL). TPP (polymer bound 3mmol/g; 10.50 g; 30.92 mmol) was added and the reaction mixture cooled in an ice bath. DTAD (7.12 g; 30.92 mmol) was added. After stirring over night at RT, the reaction mixture was filtered and the THF was evaporated. The residue was poured on water and extracted several times with EtOAc. The combined organic layers were washed with brine, dried over Na₂SO₄, filtered off and concentrated under reduced pressure. The residue was purified by column chromatography (silica gel; DCM/MeOH, 100/0 up to 95/5) to provide the product.

$C_9H_8BrN_3O_2S$	(M=302.2 g/mol)
ESI-MS:	302 [M+H] ⁺
R _t (HPLC):	0.67 min (method A)
Yield:	4.03 g; 56%
¹ H NMR (400 MHz	z, DMSO- <i>d</i> 6) δ ppm: 3.16 (s, 3H), 5.10 (s, 2H), 5.76 (d, $J = 7.86$ Hz,
1H), 7.73 (s, 1H), 7	1.84 (d, J = 7.86 Hz, 1H).

The intermediate compounds I.2 and I.3 shown in the table below were prepared using procedures analogous to those described for intermediate I.1 using appropriate starting materials. As is appreciated by those skilled in the art, these analogous examples may involve variations in general reaction conditions.

Int.	Starting materials	Structure	Reaction conditions (deviation from general procedure)
I.2	3-methyl-1,2,3,4- tetrahydropyrimidine-2,4-dione and (5-chloro-1,3-thiazol-2- yl)methanol		residue dissolved in DMF; left for 2 days; precipitate filtered off
I.3	5-methyl-1,2,3,4- tetrahydropyrimidine-2,4-dione and (2-bromo-1,3-thiazol-5- yl)methanol		unbound TPP

Int.	ESI-MS	HPLC retention time (method)	1H NMR
I.2	258	0.78 min (B)	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ ppm: 3.16 (s, 3 H), 5.21 (s, 2 H), 5.78 (d, <i>J</i> = 7.86 Hz, 1 H), 7.79 (s, 1 H), 7.82 (d, <i>J</i> = 7.86 Hz, 1 H).
I.3	302	0.74 min (A)	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ ppm: 1.75 (d, <i>J</i> = 1.01 Hz, 3 H), 5.00 (s, 2 H), 7.66 (d, <i>J</i> = 1.14 Hz, 1 H), 7.71 (s, 1 H), 11.40 (s, 1 H).

Intermediate I.4

1-[(5-bromo-1,3,4-thiadiazol-2-yl) methyl] -3-methyl-1,2,3,4-tetrahydropyrimidine-2,4dione



To a stirred mixture of (5-bromo-1,3,4-thiadiazol-2-yl) methanol (0.20 g; 1.03 mmol) and TEA (0.17 mL; 1.23 mmol) in DCM (5 mL) methanesulfonyl chloride (0.10 mL; 1.23 mmol) was added dropwise. After stirring for 1 h at RT, another methanesulfonyl chloride (0.04 mL) was added and stirred for a further hour. The reaction mixture was diluted with DCM and water. The organic layer was separated through a phase separator cartridge and evaporated. The residue was taken up in DMF (3 mL), 3-methyl-1,2,3,4- tetrahydropyrimidine-2,4-dione (0.10 g; 0.79 mmol) and K₂CO₃ (0.28 g; 2.03 mmol) were added. The reaction mixture was attired at RT overnight, filtered and purified by column chromatography (reversed phase; Sunfire C18; water/ACN/TFA) to provide the product.

 $C_8H_7BrN_4O_2S$ (M=303.1 g/mol)ESI-MS:303/305 (Br) [M+H] + R_t (HPLC):0.71 min (method A)Yield:0.14 g; 58%

Intermediate I.5

1-[(5-Bromo-1,2-thiazol-3-yl) methyl] -3-methyl-1,2,3,4-tetrahydropyrimidine-2,4-dione



3-Methyl-1,2,3,4-tetrahydropyrimidine-2,4-dione (0.04 g; 0.32 mmol), 5-bromo-3-(bromomethyl) -1,2-thiazole (0.09 mL; 0.35 mmol) and K_2CO_3 (0.10 g; 0.70 mmol) in dry DMF (2 mL) were stirred at RT for 2 h. The reaction mixture was partitioned between

EtOAc and water. The organic layer was washed with brine and water, dried over Na₂SO₄, filtered and concentrated under reduced pressure.

 $C_9H_8BrN_3O_2S$ (M=302.2 g/mol)

ESI-MS: 302 [M+H] ⁺

 R_t (HPLC): 0.71 min (method A)

Yield: 0.09 g; 97%

¹H NMR (400 MHz, DMSO-*d*₆) δ ppm: 3.15 (s, 3 H), 5.07 (s, 2 H), 5.76 (d, *J* = 7.86 Hz, 1 H), 7.56 (s, 1 H), 7.77 (d, *J* = 7.86 Hz, 1 H).

2.2 Intermediate II

[3-(Benzyloxy)-2,4-difluorophenyl]boronic acid



2-(Benzyloxy) -1,3-difluorobenzene (0.50 g; 2.27 mmol) in THF (10mL) was cooled to -78°C. n-BuLi (2.5 mol/L in THF; 1.36 mL; 3.63 mmol) was added dropwise and stirred for 50 minutes in the cold. Triisopropyl borate (0.73 mL; 3.63 mmol) was added dropwise stirred for further 10min at -78°C and then further 30minutes without cooling. The reaction was quenched with HCl (aq. solution; 4 mol/L; 5 mL) and stirred for 10minutes. The mixture was poured on water and extracted several times with EtOAc. The combined organic layers were dried over MgSO₄, filtered off and concentrated under reduced pressure. The residue was purified by column chromatography (silica gel; DCM/MeOH with few AcOH, 100/0 up to 96/4) to provide the product.

$C_{13}H_{11}BF_2O_3$	(M=264.0 g/mol)
ESI-MS:	263 [M-H] ⁻
R _t (HPLC):	0.75 min (method B)
Yield:	0.25 g; 42 %

¹H NMR (400 MHz, DMSO- d_6) δ ppm: 5.12 (s, 2 H), 7.05 (ddd, J = 10.27, 8.62, 1.39 Hz, 1 H), 7.25 (dt, J = 8.40, 6.57 Hz, 1 H), 7.30-7.56 (m, 5 H), 8.23 (br s, 2 H).

2.3 Intermediate III

Intermediate III.1

(2,4-Difluoro-3-hydroxyphenyl)boronic acid



Intermediate II.1 (0.10 g; 0.38 mmol) in a mixture of THF and MeOH (each 3 mL) was hydrogenated in a parr apparatus using Pd/C 10% (0.02 g) at RT for 3 h. The reaction mixture was concentrated under reduced pressure. The residue was purified by column chromatography (reversed phase; Sunfire C18; water/ACN/TFA) to provide the product.

 $C_6H_5BF_2O_3$ (M=173.9 g/mol)ESI-MS:173 [M-H] - R_t (HPLC):0.07 min (method B)Yield:0.05 g; 77 %¹H NMR (400 MHz, DMSO- d_6) δ ppm: 6.90 - 7.00 (m, 2 H), 8.11 (br s, 2 H), 9.81 (br s, 1 H).

2.4 Intermediate IV

Intermediate IV.1 (general procedure)

2,4-Difluoro-3-[(4-methoxyphenyl)methoxy]benzaldehyde



1-(Chloromethyl) -4-methoxybenzene (0.19 mL; 1.39 mmol) added to a mixture of 2,4difluoro-3-hydroxybenzaldehyde (0.20 g; 1.27 mmol) and K₂CO₃ (0.27 g; 1.94 mmol) in ACN (5 mL). After stirring at 60°C for 3 h, the reaction mixture was filtered and diluted with DMF/water/TFA and purified by column chromatography (reversed phase; Sunfire C18; water/ACN/TFA). ACN was evaporated in the desired fractions and the product was partitioned between DCM and water. The organic layer was dried over MgSO₄, filtered and concentrated under reduced pressure.

$C_{15}H_{12}F_2O_3$	(M=278.3 g/mol)
ESI-MS:	277 [M-H] ⁻
R _t (HPLC):	1.12 min (method A)
Yield:	0.22 g; 63 %
	z, DMSO- <i>d</i> ₆) δ ppm: 3.75 (s, 3 H), 5.14 (s, 2 H), 6.93 (d, <i>J</i> = 8.74 Hz, 2 3 H), 7.59 (ddd, <i>J</i> = 8.81, 7.41, 6.08 Hz, 1 H), 10.11 (s, 1 H).

The intermediate compounds IV.2 and IV.4 shown in the table below were prepared using procedures analogous to those described for intermediate IV.1 using appropriate starting materials. As is appreciated by those skilled in the art, these analogous examples may involve variations in general reaction conditions.

Int.	Starting materials	Structure	Reaction conditions (deviation
			from general procedure)
IV.2	Intermediate VII.1	\sim μ	stirred for 30 min at 60°C and for
	and	<u>`</u> ~~~	lh at 70°C
	4-methoxybenzyl		
	chloride		
	cinonae	F_	

IV.3	3-bromo-2,6- difluorophenol and 4-methoxybenzyl chloride	P P P P P P P P P P P P P P P P P P P	stirred for 2 h at 70°C
IV.4	Intermediate VII.2 and 4-methoxybenzyl chloride		stirred for 6.5 h at 80°C; purification by evaporating and treating with water to create precipitate

Int.	ESI-	HPLC retention	¹ H NMR
	MS	time (method)	
IV.2	390	1.20 min	
		(A)	
IV.3			
IV.4	365	1.02 (A)	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ ppm: 3.75 (s, 3 H), 4.94 (s, 2 H), 5.17 (s, 2 H), 6.30 (br s, 1 H), 6.94 (d, J = 8.62 Hz, 2 H), 7.29 - 7.40 (m, 3 H), 7.92 (ddd, J = 8.87, 7.73, 5.96 Hz, 1 H).

2.5 Intermediate V

Intermediate V.1

({2,4-Difluoro-3-[(4-methoxyphenyl)methoxy]phenyl}methylidene)hydroxylamine



Intermediate IV.1 (0.22 g; 0.79 mmol) and NaOAc (0.08 g; 1.03 mmol) in MeOH (8 mL) and water (3 mL) were stirred at RT. Hydroxylamine hydrochloride (0.04 mL; 1.03 mmol) was added and the reaction mixture heated up to 80°C for 1.5 h. The reaction mixture was concentrated under reduced pressure and taken up in water/EtOAc. The aqueous layer was

extracted several times with EtOAc. The combined organic layers were dried over MgSO₄, filtered and concentrated under reduced pressure.

 $C_{15}H_{13}F_{2}NO_{3}$ (M=293.3 g/mol)ESI-MS:294 [M+H] + R_{t} (HPLC):1.06 min (method A)Yield:0.25 g; quantitative¹H NMR (400 MHz, DMSO- d_{6}) δ ppm: 3.75 (s, 3 H), 5.10 (s, 2 H), 6.90 - 6.96 (m, 2 H), 7.10- 7.19 (m, 1 H), 7.31 - 7.37 (m, 2 H), 7.42 (ddd, J = 8.81, 7.79, 6.08 Hz, 1 H), 8.14 (s, 1 H),11.59 (s, 1 H).

2.6 Intermediate VI

Intermediate VI.1

1-[(3-{2,4-Difluoro-3-[(4-methoxyphenyl)methoxy]phenyl}-1,2-oxazol-5-yl)methyl]-3methyl-1,2,3,4-tetrahydropyrimidine-2,4-dione



Intermediate V.1 (0.10 g; 0.34 mmol), 3-methyl-1-(prop-2-yn-1-yl)-1,2,3,4tetrahydropyrimidine-2,4-dione (0.06 g; 0.34 mmol) and TEA (0.01 mL; 0.03 mmol) in *t*-BuOH (4 mL) and water (4 mL) were stirred at RT. NaOCl (0.45 mL; 0.58 mmol) was added. After stirring for 2.5 h at RT, EtOAc and water were added. The organic layer was separated, dried over MgSO₄, filtered and concentrated under reduced pressure. The residue was purified by column chromatography (reversed phase; Sunfire C18; water/ACN/TFA) to provide the product.

$C_{23}H_{19}F_2N_3O_5$	(M=455.4 g/mol)
ESI-MS:	456 [M+H] ⁺
R _t (HPLC):	1.07 min (method A)
Yield:	0.02 g; 12%

¹H NMR (400 MHz, DMSO- d_6) δ ppm: 3.17 (s, 3 H), 3.75 (s, 3 H), 5.14 (s, 2 H), 5.20 (s, 2 H), 5.81 (d, J = 7.86 Hz, 1 H), 6.89 (d, J = 2.66 Hz, 1 H), 6.93 (d, J = 8.74 Hz, 2 H), 7.26 (td, J = 9.57, 1.77 Hz, 1 H), 7.35 (d, J = 8.62 Hz, 2 H), 7.56 (ddd, J = 8.90, 7.83, 5.96 Hz, 1 H), 7.86 (d, J = 7.98 Hz, 1 H).

Intermediate VI.2 (general procedure)

1-[(5-{2,4-Difluoro-3-[(4-methoxyphenyl)methoxy]phenyl}-1,2-oxazol-3-yl)methyl]-3methyl-1,2,3,4-tetrahydropyrimidine-2,4-dione



Intermediate IX.1 (0.10 g; 0.03 mmol), 3-methyl-1,2,3,4-tetrahydropyrimidine-2,4-dione (0.01 g; 0.05 mmol) and K_2CO_3 (0.01 g; 0.08 mmol) in DMF (2 mL) were stirred for 1 h at RT and for 30 min at 50°C. The reaction mixture was filtered and purified by column chromatography (reversed phase; Sunfire C18; water/ACN/TFA) to provide the product.

$C_{23}H_{19}F_2N_3O_5$	(M=455.4 g/mol)
ESI-MS:	456 [M+H] ⁺
R _t (HPLC):	1.07 min (method A)
Yield:	0.01 g; 53%

The following intermediate compound VI.3 below was prepared using procedures analogous to those described for the intermediate VI.2 using appropriate starting materials. As is appreciated by those skilled in the art, these analogous examples may involve variations in general reaction conditions.

Int.	Starting materials	Structure
VI.3	Intermediate IX.2	0
	and	
	1,2,3,4-tetrahydropyrimidine-2,4-	
	dione	
		s_N
		F
		O^

Int.	ESI- MS	HPLC retention	¹ H NMR
		time	
		(method)	
VI.3	458	1.01 min	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ ppm: 3.73 (s, 3 H), 5.11
		(A)	(s, 2 H), 5.24 (d, <i>J</i> = 3.42 Hz, 2 H), 5.86 (d, <i>J</i> = 7.86 Hz, 1
			H), 6.91 (d, J = 8.36 Hz, 2 H), 7.18 - 7.27 (m, 1 H), 7.33
			(d, J = 8.49 Hz, 2 H), 7.84 (tt, J = 8.46, 5.67 Hz, 1 H), 7.90
			- 7.99 (m, 1 H), 8.03 (d, <i>J</i> = 2.03 Hz, 1 H).

2.7 Intermediate VII

Intermediate VII.1

Ethyl 5-(2,4-difluoro-3-hydroxyphenyl)-1,2-oxazole-3-carboxylate



Ethyl 5-chloro-1,2-oxazole-3-carboxylate (0.15 g; 0.85 mmol), intermediate III.1 (0.22 g; 1.28 mmol) and K_3PO_4 (0.36 g; 1.71 mmol) were dissolved in dioxane (3 mL). The reaction

mixture was flushed with nitrogen. Pd-Peppsi 2Me-Ipent Cl (0.04 g; 0.04 mmol) was added. After stirring for 20 min at 65°C, the reaction mixture was diluted with DMF/water and purified by column chromatography (reversed phase; Sunfire C18; water/ACN/TFA) to provide the product.

 $C_{12}H_9F_2NO_4$ (M=269.2 g/mol)ESI-MS:270 [M+H] + R_t (HPLC):0.98 min (method A)Yield:0.03 g; 11%

¹H NMR (400 MHz, DMSO- d_6) δ ppm: 1.35 (t, J = 7.16 Hz, 3 H), 4.40 (q, J = 7.10 Hz, 2 H), 7.16 (d, J = 2.91 Hz, 1 H), 7.25 (td, J = 9.57, 1.77 Hz, 1 H), 7.43 (ddd, J = 8.84, 7.64, 5.70 Hz, 1 H), 10.77 (br s, 1 H).

Intermediate VII.2 2,6-Difluoro-3-[5-(hydroxymethyl)-1,3,4-thiadiazol-2-yl]phenol



The reaction was performed under nitrogen atmosphere. (5-Bromo-1,3,4-thiadiazol-2-yl) methanol (4.00 g; 20.51 mmol), intermediate III.1 (4.60 g; 24.61 mmol) and Na₂CO₃ (5.43 g; 51.27 mmol) were dissolved in EtOH (50mL) and water (10 mL). Pd-Peppsi 2Me-Ipent Cl (0.86 g; 1.03 mmol) was added and the mixture stirred at 80° overnight. The reaction mixture was filtered and the filtrate concentrated under reduced pressure. The residue was taken up in water and the resulting precipitate filtered off to provide the product.

$C_9H_6F_2N_2O_2S$	(M=244.2 g/mol)
ESI-MS:	245 [M+H] ⁺
R _t (HPLC):	0.72min (method A)
Yield:	2.66 g; 53%

¹H NMR (400 MHz, DMSO-*d*₆) δ ppm: 4.93 (d, *J* = 4.69 Hz, 2 H), 6.27 (br t, *J* = 5.39 Hz, 1 H), 7.19 - 7.34 (m, 1 H), 7.65 (ddd, *J* = 8.84, 7.57, 5.89 Hz, 1 H), 10.75 (s, 1 H).

2.8 Intermediate VIII

Intermediate VIII.1

(5-{2,4-Difluoro-3-[(4-methoxyphenyl)methoxy]phenyl}-1,2-oxazol-3-yl)methanol



Intermediate IV.2 (0.03 g; 0.07 mmol) in THF (3 mL) was cooled to -20°C. LiAlH₄ (solution in THF; 1 mol/L; 0.05 mL; 0.05 mol/L) was added dropwise and stirred for 30 min in the cold. Two drops of water and NaOH (aq. solution; 4 mol/L) were added and stirred for further 20 min at RT. The reaction mixture was filtered, diluted with watere and purified by column chromatography (reversed phase; Sunfire C18; water/ACN/TFA) to provide the product.

$C_{18}H_{15}F_2NO_4 \\$	(M=347.3 g/mol)
ESI-MS:	348 [M+H] +
R _t (HPLC):	1.04 min (method A)
Yield:	0.02 g; 70%

¹H NMR (400 MHz, DMSO-*d*₆) δ ppm: 3.75 (s, 3 H), 4.57 (d, *J* = 6.08 Hz, 2 H), 5.16 (s, 2 H), 5.55 (t, *J* = 6.02 Hz, 1 H), 6.82 (d, *J* = 3.55 Hz, 1 H), 6.93 (d, *J* = 8.74 Hz, 2 H), 7.30 (ddd, *J* = 10.36, 8.90, 1.90 Hz, 1 H), 7.36 (d, *J* = 8.62 Hz, 2 H), 7.63 (ddd, *J* = 8.90, 7.83, 5.83 Hz, 1 H).

2.9 Intermediate IX

Intermediate IX.1

3-(Chloromethyl)-5-{2,4-difluoro-3-[(4-methoxyphenyl)methoxy]phenyl}-1,2-oxazole



Intermediate VIII.1 (0.02 g; 0.024mmol), TPP (0.01 g; 0.04 mmol) and carbon tetrachloride (0.02 mL; 0.22 mmol) in ACN (2 mL) were stirred for 2.67 h at 80°C. The reaction mixture was concentrated under reduced pressure. The residue was used in the further reaction as crude product.

$C_{18}H_{14}ClF_2NO_3$	(M=365.8 g/mol)
ESI-MS:	366 [M+H] ⁺
R _t (HPLC):	1.20 min (method A)
Yield:	0.01 g; 70% (determined by HPLC-MS)

Intermediate IX.2 (general procedure)

5-(Chloromethyl)-2-{2,4-difluoro-3-[(4-methoxyphenyl)methoxy]phenyl}-1,3-thiazole



Intermediate XI.1 (0.99 g; 2.72 mmol) and TEA (0.60 mL; 4.33 mmol) in DCM (10 mL) were stirred in an ice bath. Methanesulfonyl chloride (0.32 mL; 4.13 mmol) was added dropwise. After stirring for 5 min in the cold, the reaction mixture was stirred at RT overnight. The reaction mixture was diluted with additional DCM and washed with diluted citric acid. The organic layer was separated with a phase separator cartridge and evaporated. $C_{18}H_{14}ClF_2NO_2S$ (M=381.8 g/mol) ESI-MS: 382 [M+H] ⁺ R_t (HPLC): 1.22 min (method A) Yield: 1.01 g; 97%

¹H NMR (400 MHz, DMSO-*d*₆) δ ppm: 3.75 (s, 3 H), 5.16 (s, 2 H), 5.18 (s, 2 H), 6.93 (d,

J = 8.74 Hz, 2 H), 7.28 (td, *J* = 9.63, 1.90 Hz, 1 H), 7.36 (d, *J* = 8.62 Hz, 2 H), 7.89 (ddd, *J* = 9.00, 8.05, 6.02 Hz, 1 H), 8.03 (d, *J* = 2.28 Hz, 1 H).

The following intermediate compound IX.3 was prepared using procedures analogous to those described for the intermediate IX.2 using appropriate starting materials. As is appreciated by those skilled in the art, these analogous examples may involve variations in general reaction conditions.

Int.	Starting materials	Structure	Reaction conditions (deviation	ESI- MS	HPLC retention time	¹ H NMR
			from		(method)	
			general procedure)			
IX.3	Intermediate	CI /	procedure)	383	1.10	
	IV.4				(A)	

2.10 Intermediate X

Intermediate X.1

2,4-Difluoro-N',3-dihydroxybenzene-1-carboximidamide



2,4-Difluoro-3-hydroxybenzonitrile (0.03 g; 0.20 mmol) in EtOH (0.5 mL) was stirred at

RT. Hydroxylamine hydrochloride (0.03 g; 0.40 mmol) and TEA (0.06 mL; 0.04 mmol) in EtOH (1mL) was added dropwise. After stirring for 20 h at RT the reaction mixture was concentrated under reduced pressure and further used as crude product. $C_7H_6F_2N_2O_2$ (M=188.1 g/mol)

2.11 Intermediate XI

Intermediate XI.1

(2-{2,4-Difluoro-3-[(4-methoxyphenyl)methoxy]phenyl}-1,3-thiazol-5-yl)methanol



Intermediate IV.3 (3.00 g; 9.11 mmol), bis(pinacolato)diboron (3.50 g; 13.78 mmol), KOAc (2.30 g; 23.44 mmol) and Pd(dppf)₂ x CH₂Cl₂ (0.75 g; 0.92 mmol) in dioxane (50 mL) were stirred overnight at 90°C. Additional bis(pinacolato)diboron (0.70 g; 2.76 mmol) was added and the reaction mixture stirred for further 2 h at 100°C. After cooling to RT, Pd(dppf)₂ x CH₂Cl₂ (0.38 g; 0.46 mmol), (2-bromo-1,3-thiazol-5-yl) methanol (1.80 g; 9.28 mmol) and Na₂CO₃ (aq. solution; 2 mol/L; 15.00 mL; 30.00 mmol) were added and the reaction mixture was stirred for 6 h at reflux. The mixture was cooled to RT, diluted with EtOAc and filtered through Celite[®]. The filtrate was concentrated under reduced pressure. The residue was partitioned between EtOAc and water. The organic layer was separated, dried over Na₂SO₄, filtered and concentrated under reduced pressure. The residue was purified by column chromatography (silica gel; cyclohexane/EtOAC 2:1 to 1:1). Et₂O was added to the evaporated fractions, stirred and filtered off. The precipitate was dried in air.

$C_{18}H_{15}F_2NO_3S$	(M=363.4 g/mol)
ESI-MS:	364 [M+H] ⁺
R _t (HPLC):	1.07 min (method A)
Yield:	1.10 g; 30%

¹H NMR (400 MHz, DMSO-*d*₆) δ ppm: 3.75 (s, 3 H), 4.75 (d, *J* = 5.70 Hz, 2 H), 5.16 (s, 2 H), 5.64 (t, *J* = 5.77 Hz, 1 H), 6.93 (d, *J* = 8.74 Hz, 2 H), 7.26 (td, *J* = 9.63, 1.77 Hz, 1 H), 7.36 (d, *J* = 8.62 Hz, 2 H), 7.79 - 7.83 (m, 1 H), 7.86 (ddd, *J* = 8.93, 8.11, 6.02 Hz, 1 H).

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2.12 Intermediate XII

Intermediate XII.1 (general procedure)

1-[(2-{2,4-Difluoro-3-[(4-methoxyphenyl)methoxy]phenyl}-1,3-thiazol-5-yl)methyl]-3ethyl-1,2,3,4-tetrahydropyrimidine-2,4-dione



Intermediate VI.3 (0.03 g; 0.05 mmol), iodoethane (0.01 mL; 0.16 mmol) and K_2CO_3 (0.03 g; 0.16 mmol) in DMF (3 mL) were stirred at 50°C for 2 h. The reaction mixture was diluted with water, filtered and purified by column chromatography (reversed phase; Sunfire C18; water/ACN/TFA) to provide the product.

$C_{24}H_{21}F_2N_3O_4S\\$	(M=485.5 g/mol)
R _t (HPLC):	1.12 min (method A)
Yield:	0.02 g; 64%

The following intermediate compounds XII.2 to XII.7 below were prepared using procedures analogous to those described for the intermediate XII.1 using appropriate starting materials. As is appreciated by those skilled in the art, these analogous examples may involve variations in general reaction conditions.

Int.	Starting materials	Structure	Reaction conditions (deviation from general procedure)
XII.2	Intermediate XIII.1 and iodoethane	O O N N S Br	stirred at 70°C; purification by extraction from water with DCM
XII.3	Intermediate I.3 and iodoethane		
XII.4	Intermediate XIII.2 and iodoethane		purification by performing precipitate using water
XII.5	Intermediate XV.1 and 1,1,1-trifluoro-2- iodoethane		

XII.6	Intermediate XV.1	0 	stepwise heating up
	and		from 50° to 100°C
	(iodomethyl)cyclopropane)	· ON N	during 3.5 h
		's	
		Br	
XII.7	Intermediate XIII.3	0 	purification by
	and		extraction between
	iodoethane		water and DCM
		s	
		Br	

Int.	ESI-MS	HPLC retention time	¹ H NMR
		(method)	
XII.2	331		
XII.3	330	0.93 min	
		(A)	
XII.4	331		¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 1.08 (t, J = 7.03 Hz, 3 H), 1.82 (d, $J = 1.14$ Hz, 3 H), 3.84 (q, $J = 7.10$ Hz, 2 H), 5.34 (s, 2 H), 7.73 (q, J = 1.14 Hz, 1 H).
XII.5	370	0.95 (A)	
XII.6	342	0.96 (A)	
XII.7		0.91 (A)	

2.13 Intermediate XIII

Intermediate XIII.1 (general procedure)

1-[(5-Bromo-1,3,4-thiadiazol-2-yl)methyl]-1,2,3,4-tetrahydropyrimidine-2,4-dione



The reaction was performed under an argon atmosphere. 1,2,3,4-Tetrahydropyrimidine-2,4dion (0.50 g; 4.46 mmol) and (E)-(trimethylsilyl N-(trimethylsilyl)ethanimidate (2.75 mL; 11.25 mmol) in ACN (15 mL) were stirred for 5 h at RT. (5-Bromo-1,3,4-thiadiazol-2yl)methyl methanesulfonat (1.40 g; 5.13 mmol) and tetrabutylazanium iodide (0.17 g ; 0.46 mmol) were added. After stirring for 6 h at 80°C and overnight at RT, 30 mL of water was slowly added. The resulting precipitate was filtered off and washed with water, ACN and Et₂O.

C₇H₅BrN₄O₂S (M=289.1 g/mol) Yield: 0.29 g; 23%

The following intermediate compounds XIII.2 to XIII.4 below were prepared using procedures analogous to those described for the intermediate XIII.1 using appropriate starting materials. As is appreciated by those skilled in the art, Zanalogous examples may involve variations in general reaction conditions.

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Int.	Starting materials	Structure
XIII.2	5-methyl-1,2,3,4-tetrahydropyrimidine-2,4-dione and (5-bromo-1,3,4-thiadiazol-2-yl)methyl methanesulfonat	HN O N S Br
XIII.3	1H,2H,3H,4H,5H,6H,7H-cyclopenta[d]pyrimidine-2,4- dione and (5-bromo-1,3,4-thiadiazol-2-yl)methyl methanesulfonat	
XIII.4	Intermediate IX.3 and 1,2,3,4-tetrahydroquinazoline-2,4-dione	

Int.	ESI-MS	HPLC	¹ H NMR
		retention time	
		(method)	
XIII.2	303/305	0.67	¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 1.76 (d, $J = 0.76$

		(A)	Hz, 3 H), 5.28 (s, 2 H), 7.67 (q, <i>J</i> = 1.01 Hz, 1 H), 11.47 (s, 1 H).
XIII.3	329	0.67 (A)	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ ppm: 1.99 (br quin, J = 7.54 Hz, 2 H), 2.53 (br t, J = 7.48 Hz, 2 H), 2.93 (br t, J = 7.60 Hz, 2 H), 5.30 (s, 2 H), 11.29 (s, 1 H).
XIII.4	509	1.00 (A)	

2.14 Intermediate XIV

Intermediate XIV.1

1-[(5-Bromo-1,3,4-thiadiazol-2-yl)methyl]-6-methyl-1,2,3,4-tetrahydropyrimidine-2,4dione



(5-Bromo-1,3,4-thiadiazol-2-yl) methyl methanesulfonat (0.20 g; 0.73 mmol) and 6methyl-1,2,3,4-tetrahydropyrimidine-2,4-dione (0.18 g; 1.46 mmol) were dissolved in DMF (5 mL), K_2CO_3 (0.25 g; 1.83 mmol) is added. After stirring for 2 h at RT, iodoethane (0.15 mL; 1.83 mmol) was added and stirred for another hour at 50°C. The reaction mixture was purified by column chromatography (reversed phase; Sunfire C18; water/ACN/TFA) to provide the product.

 $C_{10}H_{11}BrN_4O_2S$ (M=331.2 g/mol)ESI-MS:331 [M+H]+ $R_f(HPLC):$ 0.83 min (method A)

Yield: 0.03 g; 14%

¹H NMR (400 MHz, DMSO-*d*₆) δ ppm: 1.08 (t, *J* = 7.03 Hz, 3 H), 2.31 (d, *J* = 0.63 Hz, 3 H), 3.82 (q, *J* = 6.97 Hz, 2 H), 5.42 (s, 2 H), 5.71 (q, *J* = 0.63 Hz, 1 H).

2.15 Intermediate XV

Intermediate XV.1

1-[(2-Bromo-1,3-thiazol-5-yl)methyl]-1,2,3,4-tetrahydropyrimidine-2,4-dione



The reaction was performed under a nitrogen atmosphere. 1,2,3,4-Tetrahydropyrimidine-2,4-dione (1.35 g; 12.04 mmol), HMDS (2.76 mL; 13.25 mmol) and chlorotrimethylsilane (0.76 mL; 6.02 mmol) in dry ACN (10 mL) were stirred for 5h at 140°C and then concentrated under reduced pressure. Half of the residue was taken up in dry ACN (22 mL), 2-bromo-5-(bromomethyl)-1,3-thiazole (2.90 g; 11.29 mmol) in ACN (13 mL) was added dropwise at 0°C. After complete addition, the mixture was heated up to 80°C and stirred overnight. After cooling to RT, the reaction mixture was diluted with NaHCO₃ (9% aq. solution) and extracted several times from EtOAc. The organic layer was separated, dried over Na₂SO₄, filtered and concentrated under reduced pressure. The residue was treated with Et₂O and filtered.

$C_8H_6BrN_3O_2S$	(M=288.1 g/mol)
ESI-MS:	288/290 [M+H] ⁺
R _t (HPLC):	0.67 min (method A)
Yield:	2.66 g; 77%

¹H NMR (400 MHz, DMSO-*d*₆) δ ppm: 5.04 (s, 2 H), 5.62 (d, *J* = 7.86 Hz, 1 H), 7.71 (s, 1 H), 7.78 (d, *J* = 7.86 Hz, 1 H), 11.41 (br s, 1 H).

2.16 Intermediate XVI

Intermediate XVI.1 (general procedure)

1-[(2-{2,4-Difluoro-3-[(4-methoxyphenyl)methoxy]phenyl}-1,3-thiazol-5-

yl)methyl]piperidin-2-one



Piperidin-2-one (0.03 g; 0.26 mmol) and sodium hydride (0.01 g ; 0.26 mmol) in DMF (2 mL) were stirred at RT for 10 min. Intermediate IX.2 (0.04 g; 0.10 mmol) was added. After stirring for 2 h, the mixture was diluted with water and MeOH, and purified by column chromatography (reversed phase; Sunfire C18; water/ACN/TFA) to provide the desired product.

The following intermediate compounds XVI.2 to XVI.4 below were prepared using procedures analogous to those described for the intermediate XVI.1 using appropriate starting materials. As is appreciated by those skilled in the art, these analogous examples may involve variations in general reaction conditions.

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Int.	Starting materials	Structure	Reaction conditions (deviation from general
XVI.2	Intermediate IX.2 and morpholin-3-one		procedure)
XVI.3	Intermediate IX.2 and 3,3-dimethyl-2,3- dihydro-1H-indol- 2-one		
XVI.4	Intermediate IX.2 and 1,2,3,4- tetrahydroquinolin- 2-one		

Int.	ESI- MS	HPLC retention time (method)
XVI.2	447	1.06
		(A)
XVI.3	507	1.24
		(A)
XVI.4	493	1.21
		(A)

2.17 Intermediate XVII

Intermediate XVII.1

tert-Butyl N-(3-{[(2-{2,4-difluoro-3-[(4-methoxyphenyl)methoxy]phenyl}-1,3-thiazol-5-yl)methyl]amino}propyl)-N-methylcarbamate



Intermediate IX.2 (0.03 g; 0.07 mmol); *tert*-butyl *N*-(3-aminopropyl)-*N*-methylcarbamate (0.01 g; 0.07 mmol) and K_2CO_3 (0.02g; 0.13 mmol) in ACN (3.84 mL) were stirred at 80°C for 2 h. Additional three equivalents of amine and base were added and stirred for further 1.4 h at 80°C. The reaction mixture was diluted with water and MeOH, and purified by column chromatography (reversed phase; Sunfire C18; water/ACN/TFA) to provide the desired product.

$C_{27}H_{33}F_2N_3O_4S$	(M=533.6 g/mol)
ESI-MS:	534 [M+H] ⁺
R _t (HPLC):	0.94 min (method A)
Yield:	0.03 g; 86%

2.18 Intermediate XVIII

Intermediate XVIII.1 (general procedure)

1-[(2-Bromo-1,3-thiazol-5-yl)methyl]-2,3-dihydro-1H-indol-2-one



Methyl 2-(2-aminophenyl) acetate hydrochloride (0.10 g; 0.50 mmol); 2-bromo-5-(bromomethyl) -1,3-thiazol (0.14 g; 0.50 mmol) and DIPA (0.21 mL; 1.24 mmol) in DCM (5 mL) were stirred at RT. The reaction mixture was concentrated under reduced pressure. The residue was taken up in DMF and water, and purified by column chromatography (reversed phase; Sunfire C18; water/ACN/TFA) to provide the desired product.

$C_{12}H_9BrN_2OS$	(M=309.2 g/mol)
ESI-MS:	309 [M+H] ⁺
R _t (HPLC):	0.96 min (method A)
Yield:	0.02 g; 14%

¹H NMR (400 MHz, DMSO-*d*₆) δ ppm: 3.61 (s, 2 H), 5.09 (s, 2 H), 7.02 (td, *J* = 7.48, 1.01 Hz, 1 H), 7.16 (d, *J* = 7.73 Hz, 1 H), 7.23 - 7.30 (m, 2 H), 7.78 (s, 1 H).
2.19 Intermediate XIX

Intermediate XIX.1 (general procedure) 1-Ethyl-7-methyl-2,3,6,7-tetrahydro-1H-purine-2,6-dione



Ethyl 4-amino-1-methyl-1H-imidazole-5-carboxylate (bromomethyl)-1,3-thiazol (0.20 g; 1.15 mmol) and isocyanatoethane (0.16 mL; 1.95 mmol) in pyridine (1 mL) were stirred for 2 h at 70°C. KOtBu (0.20 g; 1.72 mmol) is added. After stirring over night at 70°C, the reaction mixture was quenched with MeOH, concentrated under reduced pressure, taken up in DMF and water, and purified by column chromatography (reversed phase; Sunfire C18; water/ACN/TFA) to provide the desired product.

$C_8H_{10}N_4O_2$	(M=194.2 g/mol)		
ESI-MS:	195 [M+H] ⁺		
R _t (HPLC):	0.28 min (method C)		
Yield:	0.10 g; 43%		
¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 1.10 (t, $J = 6.97$ Hz, 3 H), 3.81 - 3.90 (m, 5 H), 7.90			
(s, 1 H), 11.77 (s, 1 H).			

The following intermediate compound was prepared using procedures analogous to those described for intermediate XIX.1 using appropriate starting materials. As is appreciated by those skilled in the art, these analogous examples may involve variations in general reaction conditions.

Int.	Starting materials	Structure	Reaction conditions (deviation from general procedure)
XIX.2	ethyl 5-amino-	0	Heating up to 105°C and used base for
	1,3-thiazole-4-		ring closure was NH4OH (5% aq.
	carboxylate	S S S S S S S S S S S S S S S S S S S	solution)
	and	н	
	isocyanatoethane		

Int.	ESI-MS	HPLC retention time (method)	¹ H NMR
XIX.2	198	0.25 (C)	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ ppm: 1.13 (t, <i>J</i> = 7.03 Hz, 3 H), 3.89 (q, <i>J</i> = 6.97 Hz, 2 H), 8.73 (s, 1 H), 12.23 (br s, 1 H).

2.20 Intermediate XX

Intermediate XX.1 (general procedure)

3-[(5-Bromo-1,3,4-thiadiazol-2-yl)methyl]-1-ethyl-7-methyl-2,3,6,7-tetrahydro-1H-purine-2,6-dione



Intermediate XIX.1 (0.10 g; 0.51 mmol), (5-bromo-1,3,4-thiadiazol-2-yl)methyl methanesulfonat (0.27 g; 0.99 mmol) and K₂CO₃ (0.21 g; 1.54 mmol) in DMF (6 mL) were

stirred at 80°C for 2.5 h. The reaction mixture was concentrated under reduced pressure, taken up in water and DCM. The organic layer was washed with water and brine, dried over Na₂SO₄, filtered and concentrated under reduced pressure to give the desired product, which was used in the next next without further purification.

$C_{11}H_{11}BrN_6O_2S$	(M=371.2 g/mol)
ESI-MS:	371 [M+H] ⁺
R _t (HPLC):	0.75 min (method A)
Yield:	0.20 g; 89%

The following intermediate compound was prepared using procedures analogous to those described for intermediate XX.1 using appropriate starting materials. As is appreciated by those skilled in the art, these analogous examples may involve variations in general reaction conditions.

Int.	Starting materials	Structure	ESI-MS	HPLC retentio n time (method	Reaction conditions (deviation from general procedure)
)	
XX.2	Intermediate	0 	374/376	0.73	purification by
	XIX.2			(A)	column
	and				chromatography
	(5-bromo-1,3,4-				(reversed phase;
	thiadiazol-2-	N			XBridge;
	yl)methyl	s – , Br			water(ACN/NH ₄ OH)
		ы Б			

Example 3: Preparation of Example Compounds

3.1 general procedure

1-{[2-(2,4-Difluoro-3-hydroxyphenyl)-1,3-thiazol-5-yl]methyl}-3-methyl-1,2,3,4-

tetrahydropyrimidine-2,4-dione (example compound 1)



Pd-Peppsi 2Me-Ipent Cl (0.06 g; 0.07 mmol) was added to a stirred mixture of Intermediate I.1 (0.40 g; 1.32 mmol), Intermediate III.1 (0.35 g; 2.01 mmol) and Cs_2CO_3 (1.10 g; 3.38 mmol) in water (1 mL) and EtOH (4 mL). After stirring for 1h at 100°C, the reaction mixture was filtered through Celite® and washed with EtOH. The filtrate was concentrated under reduced pressure. The residue was taken up in DCN and water. The organic layer was evaporated and the resulting precipitate was filtered off, washed with water, MeOH and Et₂O to give the desired product.

$C_{15}H_{11}F_2N_3O_3S$	(M=351.3 g/mol)
ESI-MS:	352 [M+H] ⁺
R _t (HPLC):	0.84 min (method A)
Yield:	0.43 g; 92%

¹H NMR (400 MHz, DMSO- d_6) δ ppm: 3.17 (s, 3 H), 5.20 (s, 2 H), 5.77 (d, J = 7.98 Hz, 1 H), 7.18 (td, J = 9.57, 1.90 Hz, 1 H), 7.60 (ddd, J = 8.93, 7.92, 5.96 Hz, 1 H), 7.90 (d, J = 7.86 Hz, 1 H), 8.00 (d, J = 2.28 Hz, 1 H), 10.51 - 10.71 (m, 1 H).

3.2 example compounds (Ex.) 2-14

The following compounds were prepared using procedures analogous to those described under 3.1 above using appropriate starting materials. As is appreciated by those skilled in the art, these analogous examples may involve variations in general reaction conditions.









Ex.	Starting materials	Reaction conditions (deviation from
		general procedure)
2	Intermediate I.1	used catalyst: PdCl ₂ (dtbpf)
	and	used base: Na ₂ CO ₃
	2,4,6-trifluoro-3-hydroxy	3 h at 80°C; purification by reversed phase (SunFire
	phenylboronic acid	C18; water/ACN/TFA)
3	Intermediates I.4	used catalyst: XPhos Pd G2
	and	used base: Na ₂ CO ₃
	III.1	10 min at 120°C (microwave); purification by reversed
		phase (SunFire C18; water/ACN/TFA)
4	Intermediates I.4+1	used catalyst: XPhos Pd G2
	and	used base: Na ₂ CO ₃
	III.1	2 h at 85°C; purification by reversed phase (SunFire
		C18; water/ACN/TFA)
5	Intermediates XII.2	purification by forming precipitate with ACN
	and	
	III. 1	
6	Intermediates XII.3	stirred for 1.75 h at 80°C; purification by reversed phase
	and	(SunFire C18; water/ACN/TFA)
	III.1	
7	Intermediates XII.4	stirred for 1 h at 90°C; purification by extraction from
	and	water with DCM
	III.1	
8	Intermediates XIV.1	stirred for 1.25 h at 80°C; purification by reversed phase
	and	(SunFire C18; water/ACN/TFA)
	III. 1	
9	Intermediates XII.5	stirred for 1.5 h at 80°C;
	and	purification by reversed phase (SunFire C18;
	III.1	water/ACN/TFA)

10	Intermediates XII.6	stirred for 1.5 h at 80°C; purification by reversed phase
	and	(SunFire C18; water/ACN/TFA)
	III.1	
11	Intermediates XVIII.1	stirred for 1.25 h at 80°C;
	and	purification by reversed phase (SunFire C18;
	III.1	water/ACN/TFA)
	111.1	water/rely/11/ry
12	Intermediates XX.1	stirred for 2 h at 80°C; purification by reversed phase
	and	(SunFire C18; water/ACN/TFA)
	III.1	
13	Intermediates XII.7	stirred for 1h at 90°C; purification by performing
	and	precipitate from ACN
	III.1	
14	Intermediates XX.2	stirred for 2 h at 80°C; purification by reversed phase
	and	(SunFire C18; water/ACN/TFA)
	III.1	

Analytical data for the example compounds 2-14 described in the tables above

Ex.	ESI-MS	HPLC	¹ H NMR
		Retention time	
		(method)	
2	370	0.51	¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 3.17 (s, 3 H),
		(C)	5.22 (s, 2 H), 5.78 (d, <i>J</i> = 7.86 Hz, 1 H), 7.37 (td,
			<i>J</i> = 10.84, 2.15 Hz, 1 H), 7.91 (d, <i>J</i> = 7.86 Hz, 1 H),
			8.05 (s, 1 H), 10.45 (s, 1 H).
3	353	0.55 min	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ ppm: 3.17 (s, 3 H),
		(D)	5.47 (s, 2 H), 5.82 (d, <i>J</i> = 7.86 Hz, 1 H), 7.19 - 7.32
			(m, 1 H), 7.58 - 7.70 (m, 1 H), 7.91 (d, <i>J</i> = 7.98 Hz, 1
			H), 10.76 (s, 1 H).
4	352	0.67 min	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ ppm: 3.17 (s, 3 H),
		(D)	5.13 (s, 2 H), 5.78 (d, <i>J</i> = 7.86 Hz, 1 H), 7.38 - 7.53
			(m, 2 H), 7.70 (s, 1 H), 7.82 (d, <i>J</i> = 7.86 Hz, 1 H),

H), 5.16 (s, 2 H), 7.13 - 7.24 (m, 1 H), 7.59 (td, $J = 8.36, 6.08$ Hz, 1 H), 7.77 (d, $J = 0.89$ Hz, 1 H), 7.99 (d, $J = 2.03$ Hz, 1 H), 10.59 (br s, 1 H).73810.90 min (A) ¹ H NMR (400 MHz, DMSO-d_6) δ ppm: 1.10 (t, $J = 7$ Hz, 3 H), 1.84 (d, $J = 1.01$ Hz, 3 H), 3.87 (q, $J = 6$ Hz, 2 H), 5.42 (s, 2 H), 7.25 (td, $J = 9.57, 1.77$ Hz, 1 7.64 (ddd, $J = 8.93, 7.54, 5.83$ Hz, 1 H), 7.79 $J = 1.14$ Hz, 1 H), 10.75 (s, 1 H).83810.89 min ¹ H NMR (400 MHz, DMSO-d_6) δ ppm: 1.09 (t, H NMR (400 MHz, DMSO-d_6) δ ppm: 1.09 (t, H), 5.51 (s, 2 H), 5.73 (s, 1 H), 7.20 - 7.32 (m, 1 H), 7.65 (ddd, $J = 8.93, 7.54, 5.83$ Hz, 1 H), 10.76 (br s, H).94200.95 min ¹ H NMR (400 MHz, DMSO-d_6) δ ppm: 4.64 (q, H), 7.12 - 7.24 (m, 1 H), 7.60 (ddd, $J = 8.90, 7.89,$ 6.02 Hz, 1 H), 7.93 - 8.09 (m, 2 H), 10.59 (br s, 1 H)103920.95 min ¹ H NMR (400 MHz, DMSO-d_6) δ ppm: 0.27 - 0.49				10.21 (s, 1 H).
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$				
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		2.5		
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	5			
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		(M-H) ⁻	(A)	
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$				H), 5.80 (d, $J = 7.86$ Hz, 1 H), 7.18 - 7.32 (m, 1 H),
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$				7.64 (ddd, <i>J</i> = 8.87, 7.54, 5.89 Hz, 1 H), 7.89 (d,
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$				<i>J</i> = 7.98 Hz, 1 H), 10.76 (br s, 1 H).
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	6	380	0.93 min	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ ppm: 1.09 (t,
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$			(A)	J = 7.03 Hz, 3 H), 1.81 (s, 3 H), 3.86 (q, J = 7.01 Hz, 2
$\begin{array}{ c c c c c c c c c c c c c c c c c c c$				H), 5.16 (s, 2 H), 7.13 - 7.24 (m, 1 H), 7.59 (td,
7 381 0.90 min ¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 1.10 (t, $J = 7$ (A) Hz, 3 H), 1.84 (d, $J = 1.01$ Hz, 3 H), 3.87 (q, $J = 6$ Hz, 2 H), 5.42 (s, 2 H), 7.25 (td, $J = 9.57$, 1.77 Hz, 1 7.64 (ddd, $J = 8.93$, 7.54, 5.83 Hz, 1 H), 7.79 $J = 1.14$ Hz, 1 H), 10.75 (s, 1 H). 8 381 0.89 min ¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 1.09 (t, $J = 7.03$ Hz, 3 H), 2.37 (s, 3 H), 3.84 (q, $J = 6.97$ Hz H), 5.51 (s, 2 H), 5.73 (s, 1 H), 7.20 - 7.32 (m, 1 H), 7.65 (ddd, $J = 8.93$, 7.54, 5.83 Hz, 1 H), 10.76 (br s, H). 9 420 0.95 min ¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 4.64 (q, H $J = 9.12$ Hz, 2 H), 5.24 (s, 2 H), 5.87 (d, $J = 7.98$ Hz H), 7.12 - 7.24 (m, 1 H), 7.60 (ddd, $J = 8.90$, 7.89, 6.02 Hz, 1 H), 7.93 - 8.09 (m, 2 H), 10.59 (br s, 1 H) (A) ¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 0.27 - 0.49 (A) 4 H), 1.06 - 1.20 (m, 1 H), 3.70 (d, $J = 7.10$ Hz, 2 H 5.21 (s, 2 H), 5.77 (d, $J = 7.86$ Hz, 1 H), 7.11 - 7.24				J = 8.36, 6.08 Hz, 1 H), 7.77 (d, $J = 0.89$ Hz, 1 H),
(A)(7.99 (d, <i>J</i> = 2.03 Hz, 1 H), 10.59 (br s, 1 H).
$\begin{array}{ c c c c c c c c c c c c c c c c c c c$	7	381	0.90 min	¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 1.10 (t, $J = 7.03$
7.64 (ddd, $J = 8.93$, 7.54, 5.83 Hz, 1 H), 7.79 $J = 1.14$ Hz, 1 H), 10.75 (s, 1 H).83810.89 min ¹ H NMR (400 MHz, DMSO-d_6) δ ppm: 1.09 (t, I = 7.03 Hz, 3 H), 2.37 (s, 3 H), 3.84 (q, $J = 6.97$ Hz H), 5.51 (s, 2 H), 5.73 (s, 1 H), 7.20 - 7.32 (m, 1 H), 7.65 (ddd, $J = 8.93$, 7.54, 5.83 Hz, 1 H), 10.76 (br s, H).94200.95 min ¹ H NMR (400 MHz, DMSO-d_6) δ ppm: 4.64 (q, I = 9.12 Hz, 2 H), 5.24 (s, 2 H), 5.87 (d, $J = 7.98$ Hz H), 7.12 - 7.24 (m, 1 H), 7.60 (ddd, $J = 8.90$, 7.89, 6.02 Hz, 1 H), 7.93 - 8.09 (m, 2 H), 10.59 (br s, 1 H)103920.95 min ¹ H NMR (400 MHz, DMSO-d_6) δ ppm: 0.27 - 0.49 (A)4H), 1.06 - 1.20 (m, 1 H), 3.70 (d, $J = 7.10$ Hz, 2 H) 5.21 (s, 2 H), 5.77 (d, $J = 7.86$ Hz, 1 H), 7.11 - 7.24			(A)	Hz, 3 H), 1.84 (d, $J = 1.01$ Hz, 3 H), 3.87 (q, $J = 6.97$
$\begin{array}{ c c c c c c c c c c c c c c c c c c c$				Hz, 2 H), 5.42 (s, 2 H), 7.25 (td, <i>J</i> = 9.57, 1.77 Hz, 1 H),
8 381 0.89 min ¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 1.09 (t, (A) J = 7.03 Hz, 3 H), 2.37 (s, 3 H), 3.84 (q, J = 6.97 Hz H), 5.51 (s, 2 H), 5.73 (s, 1 H), 7.20 - 7.32 (m, 1 H), 7.65 (ddd, J = 8.93, 7.54, 5.83 Hz, 1 H), 10.76 (br s, H). 9 420 0.95 min ¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 4.64 (q, J = 9.12 Hz, 2 H), 5.24 (s, 2 H), 5.87 (d, J = 7.98 Hz H), 7.12 - 7.24 (m, 1 H), 7.60 (ddd, J = 8.90, 7.89, 6.02 Hz, 1 H), 7.93 - 8.09 (m, 2 H), 10.59 (br s, 1 H) 10 392 0.95 min ¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 0.27 - 0.49 (A) ¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 0.27 - 0.49 (A) ¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 0.27 - 0.49 (A) ¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 0.27 - 0.49 (A) ¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 0.27 - 0.49 (A) ¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 0.27 - 0.49 (A) ¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 0.27 - 0.49 (A) ¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 0.27 - 0.49 (A) ¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 0.27 - 0.49				7.64 (ddd, $J = 8.93$, 7.54, 5.83 Hz, 1 H), 7.79 (q,
(A) $J = 7.03 \text{ Hz}, 3 \text{ H}), 2.37 (s, 3 \text{ H}), 3.84 (q, J = 6.97 \text{ Hz}H), 5.51 (s, 2 H), 5.73 (s, 1 H), 7.20 - 7.32 (m, 1 H),7.65 (ddd, J = 8.93, 7.54, 5.83 \text{ Hz}, 1 \text{ H}), 10.76 (br s,H).94200.95 min1H NMR (400 MHz, DMSO-d_6) \delta ppm: 4.64 (q,J = 9.12 Hz, 2 H), 5.24 (s, 2 H), 5.87 (d, J = 7.98 \text{ Hz}H), 7.12 - 7.24 (m, 1 H), 7.60 (ddd, J = 8.90, 7.89,6.02 Hz, 1 H), 7.93 - 8.09 (m, 2 H), 10.59 (br s, 1 H)103920.95 min1H NMR (400 MHz, DMSO-d_6) \delta ppm: 0.27 - 0.494 H), 1.06 - 1.20 (m, 1 H), 3.70 (d, J = 7.10 \text{ Hz}, 2 \text{ H}5.21 (s, 2 H), 5.77 (d, J = 7.86 \text{ Hz}, 1 \text{ H}), 7.11 - 7.24$	0	201	0.80 min	
H), 5.51 (s, 2 H), 5.73 (s, 1 H), 7.20 - 7.32 (m, 1 H), 7.65 (ddd, $J = 8.93$, 7.54, 5.83 Hz, 1 H), 10.76 (br s, H).94200.95 min ¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 4.64 (q, J = 9.12 Hz, 2 H), 5.24 (s, 2 H), 5.87 (d, $J = 7.98$ Hz H), 7.12 - 7.24 (m, 1 H), 7.60 (ddd, $J = 8.90$, 7.89, 6.02 Hz, 1 H), 7.93 - 8.09 (m, 2 H), 10.59 (br s, 1 H)103920.95 min ¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 0.27 - 0.49 (A)4H), 1.06 - 1.20 (m, 1 H), 3.70 (d, $J = 7.10$ Hz, 2 H)5.21 (s, 2 H), 5.77 (d, $J = 7.86$ Hz, 1 H), 7.11 - 7.24	8	381		
94200.95 min 1 H NMR (400 MHz, DMSO- d_{6}) δ ppm: 4.64 (q, H).94200.95 min 1 H NMR (400 MHz, DMSO- d_{6}) δ ppm: 4.64 (q, J = 9.12 Hz, 2 H), 5.24 (s, 2 H), 5.87 (d, J = 7.98 Hz H), 7.12 - 7.24 (m, 1 H), 7.60 (ddd, J = 8.90, 7.89, 6.02 Hz, 1 H), 7.93 - 8.09 (m, 2 H), 10.59 (br s, 1 H)103920.95 min 1 H NMR (400 MHz, DMSO- d_{6}) δ ppm: 0.27 - 0.49 f S.21 (s, 2 H), 5.77 (d, J = 7.86 Hz, 1 H), 7.11 - 7.24			(A)	
94200.95 min (A) 1 H NMR (400 MHz, DMSO- d_{6}) δ ppm: 4.64 (q, J = 9.12 Hz, 2 H), 5.24 (s, 2 H), 5.87 (d, J = 7.98 Hz H), 7.12 - 7.24 (m, 1 H), 7.60 (ddd, J = 8.90, 7.89, 6.02 Hz, 1 H), 7.93 - 8.09 (m, 2 H), 10.59 (br s, 1 H)103920.95 min (A) 1 H NMR (400 MHz, DMSO- d_{6}) δ ppm: 0.27 - 0.49 5.21 (s, 2 H), 5.77 (d, J = 7.86 Hz, 1 H), 7.11 - 7.24				
9420 0.95 min ^{1}H NMR (400 MHz, DMSO- d_{6}) δ ppm: 4.64 (q, (A)(A) $J = 9.12 \text{ Hz}, 2 \text{ H}$), 5.24 (s, 2 H), 5.87 (d, $J = 7.98 \text{ Hz}$ H), $7.12 - 7.24$ (m, 1 H), 7.60 (ddd, $J = 8.90, 7.89,$ $6.02 \text{ Hz}, 1 \text{ H}$), $7.93 - 8.09$ (m, 2 H), 10.59 (br s, 1 H 10392 0.95 min ^{1}H NMR (400 MHz, DMSO- d_{6}) δ ppm: $0.27 - 0.49$ (A)(A) 4 H), $1.06 - 1.20$ (m, 1 H), 3.70 (d, $J = 7.10 \text{ Hz}, 2 \text{ Hz}$ 5.21 (s, 2 H), 5.77 (d, $J = 7.86 \text{ Hz}, 1 \text{ H}$), $7.11 - 7.24$				
(A) $J = 9.12 \text{ Hz}, 2 \text{ H}, 5.24 \text{ (s, 2 H)}, 5.87 \text{ (d, } J = 7.98 \text{ Hz} \text{ H}, 7.12 - 7.24 \text{ (m, 1 H)}, 7.60 \text{ (ddd, } J = 8.90, 7.89, 6.02 \text{ Hz}, 1 \text{ H}), 7.93 - 8.09 \text{ (m, 2 H)}, 10.59 \text{ (br s, 1 H} \text{ H} \text{ H} \text{ (A)}$ $10 392 0.95 \text{ min}$ $(A) ^{1}\text{H NMR} \text{ (400 MHz, DMSO-} d_6 \text{) } \delta \text{ ppm: } 0.27 - 0.49 \text{ H} \text{ (A)}$ $4 \text{ H}, 1.06 - 1.20 \text{ (m, 1 H)}, 3.70 \text{ (d, } J = 7.10 \text{ Hz}, 2 \text{ H} \text{ (s, 2 H)}, 5.21 \text{ (s, 2 H)}, 5.77 \text{ (d, } J = 7.86 \text{ Hz}, 1 \text{ H}), 7.11 - 7.24 \text{ (m, 1 H)}, 7.11 - 7.$,
H), 7.12 - 7.24 (m, 1 H), 7.60 (ddd, $J = 8.90, 7.89,$ 6.02 Hz, 1 H), 7.93 - 8.09 (m, 2 H), 10.59 (br s, 1 H)103920.95 min ¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 0.27 - 0.49 (A)(A)4 H), 1.06 - 1.20 (m, 1 H), 3.70 (d, $J = 7.10$ Hz, 2 H 5.21 (s, 2 H), 5.77 (d, $J = 7.86$ Hz, 1 H), 7.11 - 7.24	9	420	0.95 min	¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 4.64 (q,
$\begin{array}{ c c c c c c c c c c c c c c c c c c c$			(A)	J = 9.12 Hz, 2 H), 5.24 (s, 2 H), 5.87 (d, $J = 7.98$ Hz, 1
10 392 0.95 min ¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 0.27 - 0.49 (A) 4 H), 1.06 - 1.20 (m, 1 H), 3.70 (d, J = 7.10 Hz, 2 H 5.21 (s, 2 H), 5.77 (d, J = 7.86 Hz, 1 H), 7.11 - 7.24				H), 7.12 - 7.24 (m, 1 H), 7.60 (ddd, <i>J</i> = 8.90, 7.89,
(A) (A) (A) $(4 \text{ H}), 1.06 - 1.20 \text{ (m, 1 H)}, 3.70 \text{ (d, } J = 7.10 \text{ Hz}, 2 \text{ Hz})$ 5.21 (s, 2 H), 5.77 (d, $J = 7.86 \text{ Hz}, 1 \text{ H}), 7.11 - 7.24$				6.02 Hz, 1 H), 7.93 - 8.09 (m, 2 H), 10.59 (br s, 1 H).
5.21 (s, 2 H), 5.77 (d, <i>J</i> = 7.86 Hz, 1 H), 7.11 - 7.24	10	392	0.95 min	¹ H NMR (400 MHz, DMSO- d_6) δ ppm: 0.27 - 0.49 (m,
			(A)	4 H), 1.06 - 1.20 (m, 1 H), 3.70 (d, <i>J</i> = 7.10 Hz, 2 H),
(m, 1 H), 7.60 (td, $J = 8.40$, 6.02 Hz, 1 H), 7.90 (d,				5.21 (s, 2 H), 5.77 (d, <i>J</i> = 7.86 Hz, 1 H), 7.11 - 7.24
				(m, 1 H), 7.60 (td, $J = 8.40$, 6.02 Hz, 1 H), 7.90 (d,
J = 7.86 Hz, 1 H), 8.00 (d, J = 2.15 Hz, 1 H), 10.59				J = 7.86 Hz, 1 H), 8.00 (d, J = 2.15 Hz, 1 H), 10.59 (s,
1 H).				1 H).

11	359	0.97 min	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ ppm: 3.64 (s, 2 H),
		(A)	5.18 (s, 2 H), 7.02 (td, J = 7.41, 1.01 Hz, 1 H), 7.11 -
			7.22 (m, 2 H), 7.21 - 7.32 (m, 2 H), 7.57 (ddd,
			J = 8.93, 7.92, 5.96 Hz, 1 H), 8.05 (d, J = 2.28 Hz, 1
			H), 10.56 (s, 1 H).
12	421	0.78 min	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ ppm: 1.14 (br t,
		(A)	J = 6.91 Hz, 3 H), 3.84 - 4.03 (m, 5 H), 5.65 (s, 2 H),
			7.23 (br t, $J = 9.38$ Hz, 1 H), 7.55 - 7.68 (m, 1 H), 8.06
			(s, 1 H).
13	407	0.86 min	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ ppm: 1.09 (t,
		(A)	J = 6.97 Hz, 3 H), 2.01 (quin, J = 7.54 Hz, 2 H), 2.60
			(br t, $J = 7.35$ Hz, 2 H), 3.03 (br t, $J = 7.60$ Hz, 2 H),
			3.85 (q, J = 6.97 Hz, 2 H), 5.44 (s, 2 H), 7.09 - 7.42
			(m, 1 H), 7.64 (ddd, <i>J</i> = 8.90, 7.51, 5.89 Hz, 1 H),
			10.76 (br s, 1 H).
14	424	0.78 min	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ ppm: 1.17 (t,
		(A)	J = 6.97 Hz, 3 H), 3.97 (q, J = 6.97 Hz, 2 H), 5.68 (s, 2
			H), 7.20 - 7.33 (m, 1 H), 7.64 (ddd, <i>J</i> = 8.84, 7.51,
			5.83 Hz, 1 H), 8.83 (s, 1 H), 10.77 (s, 1 H).

3.3 example compound 15

1-{[5-(2,4-Difluoro-3-hydroxyphenyl)-1,3-thiazol-2-yl]methyl}-3-methyl-1,2,3,4-tetrahydropyrimidine-2,4-dione



3-Bromo-2,6-difluorophenol (0.04 g; 0.18mmol), bis(pinacolato)diboron (0.05 g; 0.18 mmol), XPhos Pd G2 (0.01 g; 0.01 mmol), XPhos (0.01 g; 0.01 mmol) and KOAc (0.04 g;

0.36 mmol) in EtOH (2 mL) were stirred for 10 min at 120°C in a microwave. Intermediate I.2 (0.03 g; 0.12 mmol), Na₂CO₃ (aq. solution; 2 mol/L; 0.18 mL; 0.36 mmol) and XPhos Pd G2 (0.01 g; 0.01 mmol) were added and stirred for another 10 min at 120°C in the microwave. The reaction mixture was diluted with DMF and purified by column chromatography (reversed phase; Sunfire C18; water/ACN/TFA) to provide the desired product.

$C_{15}H_{11}F_2N_3O_3S$	(M=351.3 g/mol)
ESI-MS:	352 [M+H] ⁺
R _t (HPLC):	0.61 min (method D)
Yield:	0.01 g; 31%

¹H NMR (400 MHz, DMSO-*d*₆) δ ppm: 3.17 (s, 3 H), 5.29 (s, 2 H), 5.80 (d, *J* = 7.86 Hz, 1 H), 7.09 - 7.25 (m, 2 H), 7.87 (d, *J*=7.86 Hz, 1 H), 8.10 (s, 1 H), 10.49 (s, 1 H).

<u>3.4 example compound 16</u>

1-{[3-(2,4-Difluoro-3-hydroxyphenyl)-1,2-oxazol-5-yl]methyl}-3-methyl-1,2,3,4-tetrahydropyrimidine-2,4-dione



Intermediate VI.1 (0.40 g; 1.32 mmol) in TFA (1.00 mL; 12.96 mmol) and DCM (3 mL) was stirred at RT for 20 min. The reaction mixture was concentrated under reduced pressure. The residue was purified by column chromatography (reversed phase; Sunfire C18; water/ACN/TFA) to provide the desired product.

$C_{15}H_{11}F_2N_3O_4$	(M=335.3 g/mol)
ESI-MS:	336 [M+H] ⁺
R _t (HPLC):	0.83 min (method A)
Yield:	0.01 g; 72%

¹H NMR (400 MHz, DMSO-*d*₆) δ ppm: 3.17 (s, 3 H), 5.20 (s, 2 H), 5.82 (d, *J* = 7.86 Hz, 1 H), 6.89 (d, *J* = 2.66 Hz, 1 H), 7.10 - 7.22 (m, 1 H), 7.28 (ddd, *J* = 8.71, 7.57, 5.89 Hz, 1 H), 7.86 (d, *J* = 7.86 Hz, 1 H), 10.57 (s, 1 H).

The following example compounds 17 to 22 were prepared using procedures analogous to those described under 3.4 above using appropriate starting materials. As is appreciated by those skilled in the art, these analogous examples may involve variations in general reaction conditions.

Ex.	Starting materials	Structure	Reaction conditions (deviation from general procedure)
17	Intermediate VI.2		
18	Intermediate XII.1		
19	Intermediate XVI.1		
20	Intermediate XVI.2		

21	Intermediate XVI.3	HO F S N	
22	Intermediate XVI.4	HO F S N	

Analytical data for the compounds described in the table above:

Ex.	ESI-MS	HPLC	¹ H NMR
		retention time	
		(method)	
17	336	0.82 min	¹ H NMR (400 MHz, DMSO- d_6) δ ppm:
		(A)	3.17 (s, 3 H), 5.10 (s, 2 H), 5.78 (d,
			<i>J</i> = 7.86 Hz, 1 H), 6.79 (d, <i>J</i> = 3.30 Hz, 1
			H), 7.07 - 7.14 (m, 2 H), 7.83 (d, <i>J</i> = 7.86
			Hz, 1 H).
18	366	0.88 min	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ ppm:
		(A)	1.09 (t, <i>J</i> = 7.03 Hz, 3 H), 3.84 (q,
			J = 7.05 Hz, 2 H), 5.20 (s, 2 H), 5.75 (d,
			J = 7.98 Hz, 1 H), 7.10 - 7.29 (m, 1 H),
			7.59 (td, $J = 8.33$, 6.02 Hz, 1 H), 7.88 (d,
			<i>J</i> = 7.98 Hz, 1 H), 7.99 (d, <i>J</i> = 2.28 Hz, 1
			H), 10.60 (br s, 1 H).
19	325	0.86 min	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ ppm:
		(A)	1.59 - 1.79 (m, 4 H), 2.27 (t, $J = 6.27$ Hz,
			2 H), 3.29 (t, <i>J</i> = 5.70 Hz, 2 H), 4.69 (s, 2
			H), 7.18 (td, <i>J</i> = 9.54, 1.84 Hz, 1 H), 7.60
			(ddd, <i>J</i> = 8.90, 7.95, 5.96 Hz, 1 H), 7.87
			(d, <i>J</i> = 2.28 Hz, 1 H), 10.56 (br s, 1 H).

20	327	0.79 min	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ ppm:
		(A)	3.34 - 3.44 (m, 2 H), 3.77 - 3.88 (m, 2 H),
			4.09 (s, 2 H), 4.77 (s, 2 H), 7.19 (td,
			<i>J</i> = 9.57, 1.77 Hz, 1 H), 7.61 (ddd,
			J = 8.93, 7.92, 5.96 Hz, 1 H), 7.91 (d,
			J = 2.41 Hz, 1 H), 10.57 (s, 1 H).
21	387	1.04 min	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ ppm:
		(A)	1.29 (s, 6 H), 5.20 (s, 2 H), 7.02 - 7.09
			(m, 1 H), 7.12 - 7.28 (m, 3 H), 7.36 (d,
			J = 7.22 Hz, 1 H), 7.56 (ddd, J = 8.93,
			7.92, 5.96 Hz, 1 H), 8.04 (d, <i>J</i> = 2.28 Hz,
			1 H), 10.55 (br s, 1 H).
22	373	0.99 min	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ ppm:
		(A)	2.65 (dd, <i>J</i> = 8.62, 6.21 Hz, 2 H), 2.83 -
			2.94 (m, 2 H), 5.37 (s, 2 H), 7.00 (td,
			<i>J</i> = 7.35, 1.01 Hz, 1 H), 7.13 - 7.19 (m, 1
			H), 7.21 - 7.27 (m, 2 H), 7.27 - 7.33 (m, 1
			H), 7.56 (ddd, <i>J</i> = 8.87, 7.98, 5.96 Hz, 1
			H), 8.00 (d, <i>J</i> = 2.28 Hz, 1 H), 10.54 (br
			s, 1 H).

3.5 example compound 23

1-{[3-(2,4-Difluoro-3-hydroxyphenyl)-1,2,4-oxadiazol-5-yl]methyl}-3-methyl-1,2,3,4-tetrahydropyrimidine-2,4-dione



A mixture of 2-(3-methyl-2,4-dioxo-1,2,3,4-tetrahydropyrimidin-1-yl)acetic acid (0.04 g; 0.20 mmol), EDC hydrochloride (0.04 g; 0.20 mmol) and HOBT (0.03 g; 0.24 mmol) in DMF (1 mL) was stirred at RT for 15 min. This mixture was added to Intermediate X.1

(0.04 g; 0.20 mmol) in DMF (1 mL). The resulting reaction mixture was stirred for 2 h at 95°C and overnight at RT. The reaction mixture was diluted with DMF/water, filtered and purified by column chromatography (reversed phase; Sunfire C18; water/ACN/TFA) to provide the desired product.

¹H NMR (400 MHz, DMSO-*d*₆) δ ppm: 3.17 (s, 3 H), 5.40 (s, 2 H), 5.87 (d, *J* = 7.98 Hz, 1 H), 7.22 (td, *J* = 9.54, 1.71 Hz, 1 H), 7.40 (ddd, *J* = 8.87, 7.35, 5.96 Hz, 1 H), 7.88 (d, *J* = 7.98 Hz, 1 H), 10.67 (s, 1 H).

3.6 example compound 24

1-{[2-(2,4-difluoro-3-hydroxyphenyl)-1,3-thiazol-5-yl]methyl}-3-methyl-1,3-diazinan-2-one



Intermediate XVII.1 (0.03 g; 0.06 mmol) and TFA (1.00 mL; 12.96 mmol) in DCM (3 mL) were stirred at RT for 25 min. The reaction mixture was concentrated under reduced pressure. The residue was dissolved in THF (3 mL) and CDI (0.01 g; 0.07 mmol) was added. After stirring at 50°C for 2 h, the reaction mixture was diluted with DMF and water, and purified by column chromatography (reversed phase; Sunfire C18; water/ACN/TFA) to provide the desired product.

$C_{15}H_{15}F_2N_3O_2S$	(M=339.4 g/mol)
ESI-MS:	340 [M+H] ⁺
R _t (HPLC):	0.86 min (method A)
Yield:	0.01 g; 52%

¹H NMR (400 MHz, DMSO- d_6) δ ppm: 1.86 (quin, J = 5.92 Hz, 2 H), 2.82 (s, 3 H), 3.22 (dt, J = 20.40, 5.89 Hz, 4 H), 4.63 (s, 2 H), 7.13 - 7.22 (m, 1 H), 7.60 (ddd, J = 8.78, 8.02, 6.02 Hz, 1 H), 7.83 (d, J = 2.28 Hz, 1 H), 10.55 (br s, 1 H).

3.7 example compound 25 (general procedure)

1-{[5-(2,4-Difluoro-3-hydroxyphenyl)-1,3,4-thiadiazol-2-yl]methyl}-3-hydroxy-3-methyl-2,3-dihydro-1H-indol-2-one



Intermediate IX.3 (0.10 g; 0.26 mmol), 3-hydroxy-3-methyl-2,3-dihydro-1H-indol-2-one (0.04 g; 0.26 mmol) and K₂CO₃ (0.09 g; 0.65 mmol) in DMF (3 mL) were stirred for 20 h at 70°C. Water was added and the reaction mixture was extracted several times with DCM. The organic layer was separated, dried and concentrated under reduced pressure. The residue was purified by column chromatography (reversed phase; Sunfire C18; water/ACN/TFA) to give the intermediate. This intermediate was taken up in DCM (1.5 mL) and TFA (1.00 mL; 49.60 mmol) and stirred overnight at RT. After concentrating under reduced pressure, the residue was taken up in ACN and purified by column chromatography (reversed phase; Sunfire C18; water/ACN/TFA) to provide the desired product.

$C_{18}H_{13}F_2N_3O_3S$	(M=389.4 g/mol)
ESI-MS:	390 [M+H] ⁺
R _t (HPLC):	0.79 min (method A)
Yield:	0.01 g; 6%

¹H NMR (400 MHz, DMSO- d_6) δ ppm: 1.44 (s, 3 H), 5.42 (s, 2 H), 6.09 (br s, 1 H), 7.09 (td, J = 7.45, 0.82 Hz, 1 H), 7.13 (d, J = 7.86 Hz, 1 H), 7.24 (td, J = 9.60, 1.84 Hz, 1 H), 7.27 - 7.35 (m, 1 H), 7.39 (dd, J = 7.35, 0.76 Hz, 1 H), 7.63 (ddd, J = 8.93, 7.54, 5.83 Hz, 1 H), 10.74 (s, 1 H).

The following example compounds 26 to 30 were prepared using procedures analogous to those described under 3.7 above using appropriate starting materials. As is appreciated by those skilled in the art, these analogous examples may involve variations in general reaction conditions.

Ex.	Starting materials	Structure	Reaction conditions (deviation from general procedure)
26	Intermediate IX.3 and		g
	6-bromo-3-		
	hydroxy-3-	ŭ	
	methyl-2,3-		
	dihydro-1H-		
	indol-2-		
	one		

Analytical data for example compound 26 described above

Ex.	ESI-MS	HPLC	¹ H NMR
		retention time	
		(method)	
26	468	0. 88 min	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ ppm: 1.43 (s, 3 H),
		(A)	5.45 (s, 2 H), 6.17 (br s, 1 H), 7.21 - 7.30 (m, 2 H), 7.31 - 7.36 (m, 1 H), 7.44 (d, <i>J</i> = 1.65 Hz, 1 H), 7.64 (ddd, <i>J</i> = 8.97, 7.57, 5.89 Hz, 1 H), 10.75 (s, 1 H).

3.7 example compound 27

1-{[2-(2,4-Difluoro-3-hydroxyphenyl)-1,3-thiazol-5-yl]methyl}-2,3-dihydro-1H-1,3-

benzodiazol-2-one



2,3-Dihydro-1H-1,3-benzodiazol-2-one (0.03 g; 0.08 mmol) and sodium hydride (55%

dispersion in mineral oil; 0.01 g ; 0.14 mmol) in THF (1 mL) were stirred at RT for 15 min. Intermediate IX.2 (0.02 g; 0.11 mmol) was added. After stirring for 6 h at 60°C, the mixture was evaporated and taken up in water and DMF to be purified by column chromatography (reversed phase; XBridge C18; water/ACN/NH₄OH). The residue was taken up in TFA (solution in DCM; 50%; 1 mL) and stirred overnight at RT. The reaction mixture was concentrated under reduced pressure, the residue was taken up in DMF and purified by column chromatography (reversed phase; SunFire C18; water/ACN/TFA).

$C_{17}H_{11}F_2N_3O_2S$	(M=359.4 g/mol)
ESI-MS:	360 [M+H] ⁺

R_t (HPLC): 0.71 min (method E)

Yield: 0.01 g; 10%

¹H NMR (400 MHz, DMSO-d6) δ ppm: 5.30 (s, 2 H), 6.97 - 7.05 (m, 3 H), 7.12 - 7.21 (m, 1 H), 7.26 - 7.31 (m, 1 H), 7.53 - 7.61 (m, 1 H), 8.05 (d, *J* = 2.28 Hz, 1 H), 10.56 (s, 1 H), 10.96 (s, 1 H).

The following compounds 28 to 29 were prepared using procedures analogous to those described under 3.5 above using appropriate starting materials. As is appreciated by those skilled in the art, these analogous examples may involve variations in general reaction conditions.

Ex.	Starting materials	Structure	Reaction conditions (deviation from general procedure)
28	Intermediate		
	IX.2	HO F	
	and		
	1-methyl-2,3-		
	dihydro-1H-1,3-		
	benzodiazol-2-		
	one		

29	Intermediate	instead of NaH
	IX.2	2.3 equivalents
	and	Cs ₂ CO ₃ are used
	2,3-dihydro-1,3-	
	benzoxazol-2-	
	one	

Analytical data for the compounds 28 and 29 described above

Ex.	ESI-MS	HPLC retention time (method)	¹ H NMR
28	374	0.78 min	¹ H NMR (400 MHz, DMSO-d6) δ ppm: 3.35
		(E)	(s, 3 H), 5.36 (s, 2 H), 7.05 - 7.11 (m, 2 H),
			7.12 - 7.21 (m, 2 H), 7.32 - 7.41 (m, 1 H), 7.56
			(ddd, J = 8.90, 7.95, 5.96 Hz, 1 H), 8.06 (d,
			<i>J</i> = 2.28 Hz, 1 H), 10.56 (s, 1 H).
29	361	0.97 min	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ ppm: 5.38 (s,
		(E)	2 H), 7.13 - 7.21 (m, 2 H), 7.25 (td, <i>J</i> = 7.76,
			1.08 Hz, 1 H), 7.36 (dd, <i>J</i> = 7.98, 0.63 Hz, 1
			H), 7.47 (dd, <i>J</i> = 7.86, 0.76 Hz, 1 H), 7.54 -
			7.62 (m, 1 H), 8.12 (d, <i>J</i> = 2.28 Hz, 1 H), 10.60
			(br s, 1 H).

3.8 example compound 30

1-{[5-(2,4-Difluoro-3-hydroxyphenyl)-1,3,4-thiadiazol-2-yl]methyl}-3-ethyl-1,2,3,4-tetrahydroquinazoline-2,4-dione



Intermediate XIII.4 (0.07 g; 0.14 mmol), iodoethane (0.02 mL; 0.21 mmol) and K₂CO₃ (0.04 g; 0.28 mmol) in DMF (1 mL) were stirred at 80°C for 2 h. The reaction mixture was concentrated under reduced pressure and purified by column chromatography (reversed phase; Sunfire C18; water/ACN/TFA). The residue was taken up in TFA (1.00 mL; 12-96 mmol) and DCM (2.5 mL). After stirring overnight at RT the mixture was purified by column chromatography (reversed phase; Sunfire C18; water/ACN/TFA) to provide the desired compound.

$C_{19}H_{14}F_2N_4O_3S$	(M=416.4 g/mol)
ESI-MS:	417 [M+H] ⁺
R _t (HPLC):	0.90 min (method A)
Yield:	0.01 g; 16%

¹H NMR (400 MHz, DMSO-*d*₆) δ ppm: 1.20 (t, *J* = 7.03 Hz, 3 H), 4.04 (q, *J* = 7.05 Hz, 2 H), 5.84 (s, 2 H), 7.24 (ddd, *J* = 10.30, 8.90, 1.84 Hz, 1 H), 7.31 - 7.37 (m, 1 H), 7.59 - 7.64 (m, 1 H), 7.65 (d, *J* = 8.24 Hz, 1 H), 7.76 - 7.81 (m, 1 H), 8.10 (dd, *J* = 7.86, 1.52 Hz, 1 H), 10.75 (br s, 1 H).

Example 4: biochemical humanHSD17B13-RapidFire MS/MS Assay.

Estradiol (Sigma, Cat# E8875), NAD (Roche, Cat# 10621650001) and recombinant humanHSD17B13 (full-length HSD17B13 (Uniprot ID Q7Z5P4-1) with C-terminal His-tag, expressed in mammalian cells and purified to homogeneity) were diluted in assay buffer (100 mM Tris, Sigma, Cat# T2319; sodium chloride, Roth, Cat# 3957.2; 0,5mM EDTA, Invitrogen, Cat# 15575020; 0,1% TCEP, Invitrogen, Cat# T2556; 0,05% BSA fraction V (protease and fatty acid free), Serva, Cat# 11945; 0,001% Tween20, Serva, Cat# 37470). Compounds were serially diluted in DMSO (Sigma, Cat# 5879) and spotted on a 384-well Microplate, PP, V-bottom (Greiner, Cat# 781280) plate by a Labcyte Echo 55x (1% DMSO in the Assay). First, 6µL/well of recombinant hHSD17B13 (1nM final) dilution was added, followed by 15min incubation at RT. Second, 6µL/well of diluted Estradiol (30µM final) and NAD (0,5mM final) was added, mixed and incubated for 4h at RT. 1µL d4-Estrone (50nM final; Sigma, Cat#489204) followed by 2,4µL Girard's Reagent P (6,5mM final; TCI, Cat# G0030) dissolved in 90% (Sigma, Cat# 34860) methanol and 10% formic acid (Merck, Cat# 33015) was added to derivatize analytes and stop the enzyme reaction. Incubation was for 12-24h at RT before adding 70µL dH2O.

The analytical sample handling was performed by a rapid-injecting RapidFire autosampler

system (Agilent, Waldbronn, Germany) coupled to a triple quadrupole mass spectrometer (Triple Quad 6500, AB Sciex Germany GmbH, Darmstadt, Germany). Liquid sample was aspirated by a vacuum pump into a 10 μ L. sample loop for 250 ms and subsequently flushed for 3000 ms onto a C18 cartridge (Agilent, Waldbronn, Germany) with the aqueous mobile phase (99.5% water, 0.49% acetic acid, 0.01% Trifluoroacetic acid, flow rate 1.5 mL/min). The solid phase extraction step retained the analyte while removing interfering matrix (e.g., buffer components). The analyte was desorbed and eluted back from the cartridge for 3000 ms with an organic mobile phase (49.75% methanol, 49,75% acetonitrile, 0.49% acetic acid, 0.01% trifluoroacetic acid, flow rate 1.25 mL/min) and flushed into the mass spectrometer for detection in MRM mode. The MRM transition for the Estrone was 404.1 < 157.1 Da (declustering potential 27V, collision energy 43 V) and for the internal standard D4 Estrone was 408.1 < 159.1 Da (declustering potential 27V, collisionenergy 43 V). Dwell time for each MRM transition was 25 ms and pause time between MRMs was 5 ms. The mass spectrometer was operated in positive ionization mode (curtain gas 35 Au, collision gas medium, ion spray voltage 4200 V, temperature 550 °C, ion source gas 1 65 Au, ion source gas 2 80 Au). While performing the back flush into the mass spectrometer, the sample loop and relevant tubing were flushed with the organic mobile phase to prevent carryover of analyte or matrix components into the next sample. Equilibration time for the system was 500 ms. To minimize carryover effects, the wash station of the RapidFire system was used to perform needle washes with pure water (100%) and pure methanol (100%) between samples. The solvent delivery setup of the RapidFire system consists of two continuously running and isocratically operating HPLC pumps (G1310A, Agilent, Waldbronn, Germany) and one binary HPLC pump channel B (G4220A, Agilent, Waldbronn, Germany). MS data processing was performed in GMSU (Alpharetta, GA, USA), and peak area ratio analyte/internal standard was reported for IC50 calculation. IC50 values were calculated using a 4-parameter non-linear regression curve fitting model (Software Megalab inhouse development). For data evaluation and calculation, the measurement of the bottom (no HSD17B13 enzyme) was set as 0% control and the measurement of the top (includes NAD, Estrone and HSD17B13) was set as 100% control. The IC50 values were calculated using the standard 4 parameter logistic regression formula: $Y = Bottom + (Top-Bottom)/(1+10^{(LogIC50-X)} \times Slope + log((Top-Bottom)/(Fifty-$ *Bottom*)-1))).

Example 5: biochemical mouseHSD17B13-RapidFire MS/MS Assay

Estradiol (Sigma, Cat# E8875), NAD (Roche, Cat# 10621650001) and recombinant mouseHSD17B13 (U-Protein Express BV, Netherlands) were diluted in assay buffer (100 mM Tris, Sigma, Cat# T2319; sodium chloride, Roth, Cat# 3957.2; 0,5mM EDTA, Invitrogen, Cat# 15575020; 0,1% TCEP, Invitrogen, Cat# T2556; 0,05% BSA fraction V (protease and fatty acid free), Serva, Cat# 11945; 0,001% Tween20, Serva, Cat# 37470). Compounds were serially diluted in DMSO (Sigma, Cat# 5879) and spotted on a 384-well Microplate, PP, V-bottom (Greiner, Cat# 781280) plate by a Labcyte Echo 55x (1% DMSO in the Assay). First, 6µL/well of recombinant mouseHSD17B13 (50nM final) dilution was added, followed by 15min incubation at RT. Second, 6µL/well of diluted Estradiol (30µM final) and NAD (0,5mM final) was added, mixed and incubated for 3h at RT. Adding 1µL d4-Estrone (50nM final; Sigma, Cat#489204) followed by 2,4µL Girard's Reagent P (6,5mM final; TCI, Cat# G0030) dissolved in 90% (Sigma, Cat# 34860) Methanol and 10% formic acid (Merck, Cat# 33015) to derivatize analytes and stop the enzyme reaction. Incubation was for 12-24h at RT before adding 70µL dH2O.

The analytical sample handling was performed by a rapid-injecting RapidFire autosampler system (Agilent, Waldbronn, Germany) coupled to a triple quadrupole mass spectrometer (Triple Quad 6500, AB Sciex Germany GmbH, Darmstadt, Germany). Liquid sample was aspirated by a vacuum pump into a 10 µL sample loop for 250 ms and subsequently flushed for 3000 ms onto a C18 cartridge (Agilent, Waldbronn, Germany) with the aqueous mobile phase (99.5% water, 0.49% acetic acid, 0.01% trifluoroacetic acid, flow rate 1.5 mL/min). The solid phase extraction step retained the analyte while removing interfering matrix (e.g., buffer components). The analyte was desorbed and eluted back from the cartridge for 3000 ms with an organic mobile phase (49.75% methanol, 49,75% acetonitrile, 0.49% acetic acid, 0.01% trifluoroacetic acid, flow rate 1.25 mL/min) and flushed into the mass spectrometer for detection in MRM mode. The MRM transition for the Estrone was $404.1 \le 157.1$ Da (declustering potential 27V, collision energy 43 V) and for the internal standard D4 Estrone was 408.1 < 159.1 Da (declustering potential 27V, collisionenergy 43 V). Dwell time for each MRM transition was 25 ms and pause time between MRMs was 5 ms. The mass spectrometer was operated in positive ionization mode (curtain gas 35 Au, collision gas medium, ion spray voltage 4200 V, temperature 550 °C, ion source gas 1 65 Au, ion source gas 2 80 Au). While performing the back flush into the mass spectrometer, the sample loop

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and relevant tubing were flushed with the organic mobile phase to prevent carryover of analyte or matrix components into the next sample. Equilibration time for the system was 500 ms. To minimize carryover effects, the wash station of the RapidFire system was used to perform needle washes with pure water (100%) and pure methanol (100%) between samples. The solvent delivery setup of the RapidFire system consisted of two continuously running and isocratically operating HPLC pumps (G1310A, Agilent, Waldbronn, Germany) and one binary HPLC pump channel B (G4220A, Agilent, Waldbronn, Germany). MS data processing was performed in GMSU (Alpharetta, GA, USA), and peak area ratio analyte/internal standard was reported for IC50 calculation. IC50 values were calculated using a 4-parameter non-linear regression curve fitting model (Software Megalab inhouse development). For data evaluation and calculation, the measurement of the bottom (no HSD17B13 enzyme) was set as 0% control and the measurement of the top (includes NAD, Estrone and HSD17B13) was set as 100% control. The IC50 values were calculated using the standard 4 parameter logistic regression formula: Y=Bottom + (Top- $Bottom)/(1+10^{((LogIC50-X) x Slope + log((Top-Bottom)/(Fifty-Bottom)-1)))).$

Example 6: cellular human HSD17B13 - RapidFire MS/MS Assay

Estradiol (Sigma, Cat# E8875) dilution and cells (clonal HEK293 cells stabily overexpressing humanHSD17B13-Myc/DDK tagged, Lakepharma) were prepared in serum free medium (DMEM, Sigma, Cat# D5796; 10% heat inactivated FBS, Gibco, Cat# 100500; 1x Glutamax, Gibco, Cat# 35050-087; 1x sodium pyruvate, Gibyo, Cat# 11360070). 25µL of a 0,4*10^6 cells/mL dilution was seeded on a 384-well Microplate (culture-plate, Perkin Elmer, Cat# 6007680) 24h prior to compound testing. Compounds were serially diluted in DMSO (Sigma, Cat# 5879) and spotted on the preseeded cell plate, by a Labcyte Echo 55x (1% DMSO in the Assay) and incubated for 30min at 37°C in a humidified incubator (rH = 95%, CO2 = 5%). Afterwards 25µL of 60µM Estradiol dilution was added to the plate and incubated for 3h at 37°C in a humidified incubator (rH = 95%, CO2 = 5%).

20μL supernatant were taken and 2.5μL d4-Estrone (50nM final; Sigma, Cat#489204) were added followed by 5μL Girard's Reagent P (6,5mM final; TCI, Cat# G0030) dissolved in 90% (Sigma, Cat# 34860) methanol and 10% formic acid (Merck, Cat# 33015) to derivatize analytes. Incubabation was for 12-24h at RT before adding 70μL dH2O.

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The analytical sample handling was performed by a rapid-injecting RapidFire autosampler system (Agilent, Waldbronn, Germany) coupled to a triple quadrupole mass spectrometer (Triple Quad 6500, AB Sciex Germany GmbH, Darmstadt, Germany). Liquid sample was aspirated by a vacuum pump into a 10 μ L. sample loop for 250 ms and subsequently flushed for 3000 ms onto a C18 cartridge (Agilent, Waldbronn, Germany) with the aqueous mobile phase (99.5% water, 0.49% acetic acid, 0.01% trifluoroacetic acid, flow rate 1.5 mL/min). The solid phase extraction step retained the analyte while removing interfering matrix (e.g., buffer components). The analyte was desorbed and eluted back from the cartridge for 3000 ms with an organic mobile phase (49.75% methanol, 49,75% acetonitrile, 0.49% acetic acid, 0.01% trifluoroacetic acid, flow rate 1.25 mL/min) and flushed into the mass spectrometer for detection in MRM mode. The MRM transition for the Estrone was 404.1 < 157.1 Da (declustering potential 27V, collision energy 43 V) and for the internal standard D4 Estrone was 408.1 < 159.1 Da (declustering potential 27V, collisionenergy 43 V). Dwell time for each MRM transition was 25 ms and pause time between MRMs is 5 ms. The mass spectrometer was operated in positive ionization mode (curtain gas 35 Au, collision gas medium, ion spray voltage 4200 V, temperature 550 °C, ion source gas 1 65 Au, ion source gas 2 80 Au). While performing the back flush into the mass spectrometer, the sample loop and relevant tubing were flushed with the organic mobile phase to prevent carryover. MS data processing was performed in GMSU (Alpharetta, GA, USA), and peak area ratio analyte/internal standard was reported for IC50 calculation. IC50 values were calculated using a 4-parameter non-linear regression curve fitting model (Software Megalab inhouse development). For data evaluation and calculation, the measurement of the bottom (cells with Estradiol and 10µM of an inhouse identified HSD17B13 inhibitor) was set as 0% control and the measurement of the top (includes cells with Estradiol) was set as 100% control. The IC50 values were calculated using the standard 4 parameter logistic regression formula: $Y = Bottom + (Top-Bottom)/(1+10^{(LogIC50-X)} x Slope + log((Top-$ Bottom)/(Fifty-Bottom)-1))).

Example 7: cellular human HSD17B13 viability assay

Estradiol (Sigma, Cat# E8875) dilution and cells (clonal HEK293 cells stably overexpressing humanHSD17B13-Myc/DDK tagged, Lakepharma) were prepared in serum free medium (DMEM, Sigma, Cat# D5796; 10% heat inactivated FBS, Gibco, Cat# 100500; 1x Glutamax, Gibco, Cat# 35050-087; 1x sodium pyruvate, Gibyo, Cat# 11360070).

25μL of a 0,4*10^6 cells/mL dilution were seeded on a 384-well Microplate (culture-plate, Perkin Elmer, Cat# 6007680) 24h prior to compound testing.

Compounds were serially diluted in DMSO (Sigma, Cat# 5879) and spotted on the preseeded cell plate, by a Labcyte Echo 55x (1% DMSO in the Assay). Incubation was for 30min at 37°C in a humidified incubator (rH = 95%, CO2 = 5%). Afterwards 25µL of 60µM Estradiol dilution were added to the plate and incubated for 3h at 37°C in a humidified incubator (rH = 95%, CO2 = 5%).

5μL CellTiter Glo 2 (Promega, Cat# G9242) were added onto the cell plate, incubated for 15min at room temperature and luminescence was measured on PHERAstar FSX (BMG Labtech, Ortenberg, Germany).

IC50 values were calculated using a 4-parameter non-linear regression curve fitting model (Software Megalab inhouse development). For data evaluation and calculation, the measurement of the bottom (no cells, with Estradiol) was set as 0% control and the measurement of the top (includes cells and Estradiol) was set as 100% control. The IC50 values were calculated using the standard 4 parameter logistic regression formula: $Y=Bottom + (Top-Bottom)/(1+10^{(LogIC50-X)*}Slope + log((Top-Bottom)/(Fifty-Bottom)-1))).$

Example 8: biochemical humanHSD17B11-RapidFire MS/MS Assay.

Estradiol (Sigma, Cat# E8875), NAD (Roche, Cat# 10621650001) and recombinant hHSD17B11 (U-Protein Express BV, Netherlands) were diluted in assay buffer (100 mM Tris, Sigma, Cat# T2319; sodium chloride, Roth, Cat# 3957.2; 0,5mM EDTA, Invitrogen, Cat# 15575020; 0,1% TCEP, Invitrogen, Cat# T2556; 0,05% BSA fraction V (protease and fatty acid free), Serva, Cat# 11945; 0,001% Tween20, Serva, Cat# 37470). Compounds were serially diluted in DMSO (Sigma, Cat# 5879) and spotted on a 384-well Microplate, PP, V-bottom (Greiner, Cat# 781280) plate by a Labcyte Echo 55x (1% DMSO in the Assay). First, 6µL/well of recombinant hHSD17B11 (35nM final) dilution was added, followed by 15min incubation at RT. Second, 6µL/well of diluted Estradiol (30µM final) and NAD (0,5mM final) were added, mixed and incubated for 4h at RT. 1µL d4-Estrone (50nM final; Sigma, Cat#489204) followed by 2,4µL Girard's Reagent P (6,5mM final; TCI, Cat# G0030) dissolved in 90% (Sigma, Cat# 34860) methanol and 10% formic acid (Merck, Cat# 33015) were added to derivatize analytes and stop the enzyme reaction. Incubation was for 12-24h at RT before adding 70µL dH2O.

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The analytical sample handling was performed by a rapid-injecting RapidFire autosampler system (Agilent, Waldbronn, Germany) coupled to a triple quadrupole mass spectrometer (Triple Quad 6500, AB Sciex Germany GmbH, Darmstadt, Germany). Liquid sample was aspirated by a vacuum pump into a 10 μ L. sample loop for 250 ms and subsequently flushed for 3000 ms onto a C18 cartridge (Agilent, Waldbronn, Germany) with the aqueous mobile phase (99.5% water, 0.49% acetic acid, 0.01% trifluoroacetic acid, flow rate 1.5 mL/min). The solid phase extraction step retained the analyte while removing interfering matrix (e.g., buffer components). The analyte was desorbed and eluted back from the cartridge for 3000 ms with an organic mobile phase (49.75% methanol, 49,75% acetonitrile, 0.49% acetic acid, 0.01% trifluoroacetic acid, flow rate 1.25 mL/min) and flushed into the mass spectrometer for detection in MRM mode. The MRM transition for the Estrone was 404.1 < 157.1 Da (declustering potential 27V, collision energy 43 V) and for the internal standard D4 Estrone was 408.1 < 159.1 Da (declustering potential 27V, collisionenergy 43 V). Dwell time for each MRM transition was 25 ms and pause time between MRMs is 5 ms. The mass spectrometer is operated in positive ionization mode (curtain gas 35 Au, collision gas medium, ion spray voltage 4200 V, temperature 550 °C, ion source gas 1 65 Au, ion source gas 2 80 Au). While performing the back flush into the mass spectrometer, the sample loop and relevant tubing were flushed with the organic mobile phase to prevent carryover of analyte or matrix components into the next sample. Equilibration time for the system was 500 ms. To minimize carryover effects, the wash station of the RapidFire system was used to perform needle washes with pure water (100%) and pure methanol (100%) between samples. The solvent delivery setup of the RapidFire system consisted of two continuously running and isocratically operating HPLC pumps (G1310A, Agilent, Waldbronn, Germany) and one binary HPLC pump channel B (G4220A, Agilent, Waldbronn, Germany). MS data processing was performed in GMSU (Alpharetta, GA, USA), and peak area ratio analyte/internal standard was reported for IC50 calculation. IC50 values were calculated using a 4-parameter non-linear regression curve fitting model (Software Megalab inhouse development). For data evaluation and calculation, the measurement of the bottom (no HSD17B13 enzyme) was set as 0% control and the measurement of the top (includes NAD, Estrone and HSD17B13) was set as 100% control. The IC50 values were calculated using the standard 4 parameter logistic regression formula: Y = Bottom + (Top- $Bottom)/(1+10^{(LogIC50-X)} \times Slope + log((Top-Bottom)/(Fifty-Bottom)-1))).$

Example 9: humanHSD17B11-RapidFire MS/MS Assay

Estradiol (Sigma, Cat# E8875), NAD (Roche, Cat# 10621650001) and recombinant hHSD17B11 (U-Protein Express BV, Netherlands) were diluted in assay buffer (100 mM Tris, Sigma, Cat# T2319; sodium chloride, Roth, Cat# 3957.2; 0,5mM EDTA, Invitrogen, Cat# 15575020; 0,1% TCEP, Invitrogen, Cat# T2556; 0,05% BSA fraction V (protease and fatty acid free), Serva, Cat# 11945; 0,001% Tween20, Serva, Cat# 37470). Compounds were serially diluted in DMSO (Sigma, Cat# 5879) and spotted on a 384-well Microplate, PP, V-bottom (Greiner, Cat# 781280) plate by a Labcyte Echo 55x (1% DMSO in the Assay). First, 6µL/well of recombinant hHSD17B11 (35nM final) dilution was added, followed by 15min incubation at RT. Second, 6µL/well of diluted Estradiol (30µM final) and NAD (0,5mM final) was added, mixed and incubated for 4h at RT. 1µL d4-Estrone (50nM final; Sigma, Cat#489204) followed by 2,4µL Girard's Reagent P (6,5mM final; TCI, Cat# G0030) dissolved in 90% (Sigma, Cat# 34860) methanol and 10% formic acid (Merck, Cat# 33015) were added to derivatize analytes and stop the enzyme reaction. Incubation was for 12-24h at RT before adding 70µL dH2O.

The analytical sample handling was performed by a rapid-injecting RapidFire autosampler system (Agilent, Waldbronn, Germany) coupled to a triple quadrupole mass spectrometer (Triple Quad 6500, AB Sciex Germany GmbH, Darmstadt, Germany). Liquid sample was aspirated by a vacuum pump into a 10 μ L. sample loop for 250 ms and subsequently flushed for 3000 ms onto a C18 cartridge (Agilent, Waldbronn, Germany) with the aqueous mobile phase (99.5% water, 0.49% acetic acid, 0.01% trifluoroacetic acid, flow rate 1.5 mL/min). The solid phase extraction step retained the analyte while removing interfering matrix (e.g., buffer components). The analyte was desorbed and eluted back from the cartridge for 3000 ms with an organic mobile phase (49.75% methanol, 49,75% acetonitrile, 0.49% acetic acid, 0.01% trifluoroacetic acid, flow rate 1.25 mL/min) and flushed into the mass spectrometer for detection in MRM mode. The MRM transition for the Estrone was 404.1 < 157.1 Da (declustering potential 27V, collision energy 43 V) and for the internal standard D4 Estrone was 408.1 < 159.1 Da (declustering potential 27V, collisionenergy 43 V). Dwell time for each MRM transition was 25 ms and pause time between MRMs is 5 ms. The mass spectrometer is operated in positive ionization mode (curtain gas 35 Au, collision gas medium, ion spray voltage 4200 V, temperature 550 °C, ion source gas 1 65 Au, ion source

gas 2 80 Au). While performing the back flush into the mass spectrometer, the sample loop and relevant tubing were flushed with the organic mobile phase to prevent carryover of analyte or matrix components into the next sample. Equilibration time for the system was 500 ms. To minimize carryover effects, the wash station of the RapidFire system was used to perform needle washes with pure water (100%) and pure methanol (100%) between samples. The solvent delivery setup of the RapidFire system consisted of two continuously running and isocratically operating HPLC pumps (G1310A, Agilent, Waldbronn, Germany) and one binary HPLC pump channel B (G4220A, Agilent, Waldbronn, Germany). MS data processing was performed in GMSU (Alpharetta, GA, USA), and peak area ratio analyte/internal standard was reported for IC50 calculation. IC50 values were calculated using a 4-parameter non-linear regression curve fitting model (Software Megalab inhouse development). For data evaluation and calculation, the measurement of the bottom (no HSD17B13 enzyme) was set as 0% control and the measurement of the top (includes NAD, Estrone and HSD17B13) was set as 100% control. The IC50 values were calculated using the standard 4 parameter logistic regression formula: Y = Bottom + (Top-Bottom)/(1+10^((LogIC50-X) x Slope + log((Top-Bottom)/(Fifty-Bottom)-1))).

Example 10: Pharmacokinetic in vitro assay of metabolic stability in liver microsomes

The metabolic degradation of a test compound was assayed at 37°C with pooled liver microsomes.

The final incubation volume of 60 μ l per time point contained TRIS buffer pH 7.6 at RT (0.1 M), magnesium chloride (5 mM), microsomal protein (0.5 - 2 mg/ml) and the test compound at a final concentration of 1 μ M.

Following a short pre-incubation period at 37°C, the reactions were initiated by addition of beta-nicotinamide adenine dinucleotide phosphate, reduced form (NADPH, 1 mM) and terminated by transferring an aliquot into solvent after different time points.

The quenched incubations were pelleted by centrifugation (10000 g, 5 min).

An aliquot of the supernatant was assayed by LC-MS/MS for the amount of parent compound remaining.

The half-life (t1/2 INVITRO) was determined by the slope of the semi-logarithmic plot of the concentration-time profile.

The intrinsic clearance (CL_INTRINSIC) was calculated by considering the amount of

protein in the incubation:

CL_INTRINSIC [µl/min/mg protein] = (Ln 2 / (half-life [min] * protein content [mg/ml])) * 1000

Example 11: Pharmacokinetic in vitro assay of metabolic stability in in human hepatocytes (HHEP assay)

An assay in human hepatocytes was performed to assess the metabolic stability of compounds. The metabolic degradation of a test compound was assayed in a human hepatocyte suspension. After recovery from cryopreservation, human hepatocytes were diluted in DMEM (supplemented with $3.5 \ \mu g$ glucagon/500 ml, $2.5 \ mg$ insulin/500 ml, $3.75 \ mg$ hydrocortison/500 ml, 5% or 50% human serum or in absence of serum) to obtain a final cell density of $1.0 \ x \ 10^6 \ cells/ml$ or $4.0 \ x \ 10^6 \ cells/ml$, depending on the metabolic turnover rate of the test compound.

After a 30 min preincubation in a cell culture incubator (37°C, 10% CO2) test compound solution was spiked into the hepatocyte suspension, resulting in a final test compound concentration of 1 μ M and a final DMSO concentration of 0.05%.

The cell suspension was incubated at 37°C (cell culture incubator, horizontal shaker) and samples were removed from the incubation after 0, 0.5, 1, 2, 4 and 6 hours. Samples were quenched with acetonitrile (containing internal standard) and pelleted by centrifugation. The supernatant was transferred to a 96-deepwell plate, and prepared for analysis of decline of parent compound by HPLC-MS/MS.

The percentage of remaining test compound was calculated using the peak area ratio (test compound/internal standard) of each incubation time point relative to the time point 0 peak area ratio. The log-transformed data were plotted versus incubation time, and the absolute value of the slope obtained by linear regression analysis was used to estimate in vitro half-life ($T_{1/2}$).

In vitro intrinsic clearance (Cl_{int}) was calculated from in vitro $T_{1/2}$ and scaled to whole liver using a hepatocellularity of 120 x 10⁶ cells/g liver, a human liver per body weight of 25.7 g liver/kg as well as in vitro incubation parameters, applying the following equation:

CL_INTRINSIC_INVIVO [ml/min/kg] = (CL_INTRINSIC [µl/min/10⁶cells] x

hepatocellularity [10⁶ cells/g liver] x liver factor [g/kg body weight]) / 1000

Hepatic in vivo blood clearance (CL) was predicted according to the well-stirred liver model

considering an average liver blood flow (QH) of 20.7 ml/min/kg:

CL [ml/min/kg] = CL_INTRINSIC_INVIVO [ml/min/kg] x hepatic blood flow [ml/min/kg]

/(CL_INTRINSIC_INVIVO [ml/min/kg] + hepatic blood flow [ml/min/kg])

Results were expressed as percentage of hepatic blood flow:

QH [%] = CL [ml/min/kg] / hepatic blood flow [ml/min/kg])

Example compound	IC ₅₀ [nM], biochemical human HSD17B13	IC ₅₀ [nM], biochemical mouse HSD17B13	IC ₅₀ [nM], cellular human HSD17B13	IC ₅₀ [nM], biochemical human HSD17B11
1	9.19	319	30.9	4210
2	26.8	2420	128	8610
3	12.2	2420	40.4	
4	4.86	446	76.4	>10300
5	25.9	2560	116	
6	11.4	1110	33.5	8630
7	624	>10300	1760	>10300
8	59.7	2090	183	263
9	3.85	52.8	17.5	6090
10	<3.09	63.1	15.3	8260
11	<3.09	27.6	12.1	5940
12	0.993	26.8	15.1	>10300
13	4.18	73.1	24	>10300
14	3.67	22.4	23.5	3540
15	3.68	13.8	39.8	3260
16	150	1620	868	5900
17	272	1350	1280	8690
18	316	1710	1180	7000
19	12.6	665	60.6	1210
20	5.72	48.1	24.3	992
21	7.39	108	93.5	6320
22	4.45	25.1	34.4	5040
23	76.8	1310	157	1740
24	15.8	204	78.4	2520
25	1.99	45.9	19.6	> 10300
26	1.26	23.2	5.13	10200
27	1.59	29.9	20.6	6640
28	12.1	235	37.4	1110
29	24.9	324	68.5	991
30	1.05	15.5	<3.23	2260

Example 12: Biological data of the example compounds 1-30

<u>Claims</u>

1. A compound having formula (I) or a salt thereof



wherein **a) Z**—— * is selected from the group consisting of





and wherein







2. The compound of claim 1, wherein the structure of the group of structures consisting of



3. The compound of claim 1 wherein Z—— * is



- **4.** The compound of claim 1 or a salt thereof selected from the group of formula (I) compounds 01-30.
- 5. The salt of a compound according to any one of claims 1 to 4 for use as a medicament.
- 6. A medicament prepared with a compound or salt thereof according to any one of claims 1 to 4.
- 7. A method for the preparation of a compound of claim 1 with general formula (I), comprising reacting compound X1, wherein Y can be Br, I or Cl, with compound X2 to give compound (I)



wherein in compounds of X1 and (I) Z, A1, A2 and A3 have the same meaning as defined in claim 1.

- The method of claim 7 taking place in an aprotic or protic solvent or a solvent mixture at a temperature between ambient temperatur and the boiling point of the solvent, preferably between 50°C and 120°C.
- **9.** A pharmaceutical composition comprising at least one compound according to one or more of the claims 1 to 4 or a pharmaceutically acceptable salt thereof and one or more pharmaceutically acceptable excipients.
- 10. The pharmaceutical composition according to claim 9 comprising a therapeutically effective amount of a compound according to any one of claims 1 to 5 in the range from 0.1 to 90 wt.-% of the composition, preferably in the range from 0.5 to 50 wt.-% of the composition, or a pharmaceutically acceptable salt thereof.
- 11. A compound according to one or more of claims 1 to 4 or a pharmaceutically acceptable salt thereof or a pharmaceutical composition according to claim 10 for use in the treatment or prevention of steatosis such as non-alcoholic fatty liver disease (NAFLD) or non-alcoholic steatohepatitis (NASH).
- 12. A compound according to any one of claims 1 to 5 for use in the treatment and/or prevention of non-alcoholic steatohepatitis, wherein said compound is administered before, after or together with at least one other active substance.

13. A compound having formula (Ia)



useful as an intermediate compound for the preparation of a compound of formula (I) according to claim 1, wherein \mathbf{Z} is as defined in claim 1 and Y can be Br, I or Cl.

14. A compound having formula (Ib)



useful as an intermediate compound for the preparation of a compound of formula (I) according to claim 1, wherein A1, A2 and A3 are as defined in claim 1.

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